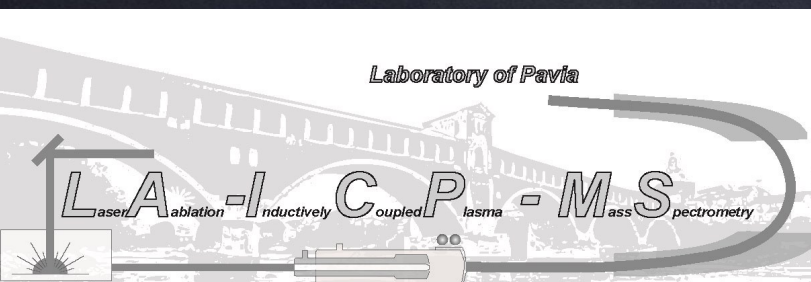


Modena, 5 novembre 2024

# “Applicazioni della Spettrometria di Massa con campionamento LASER nelle Geoscienze”

Alberto Zanetti

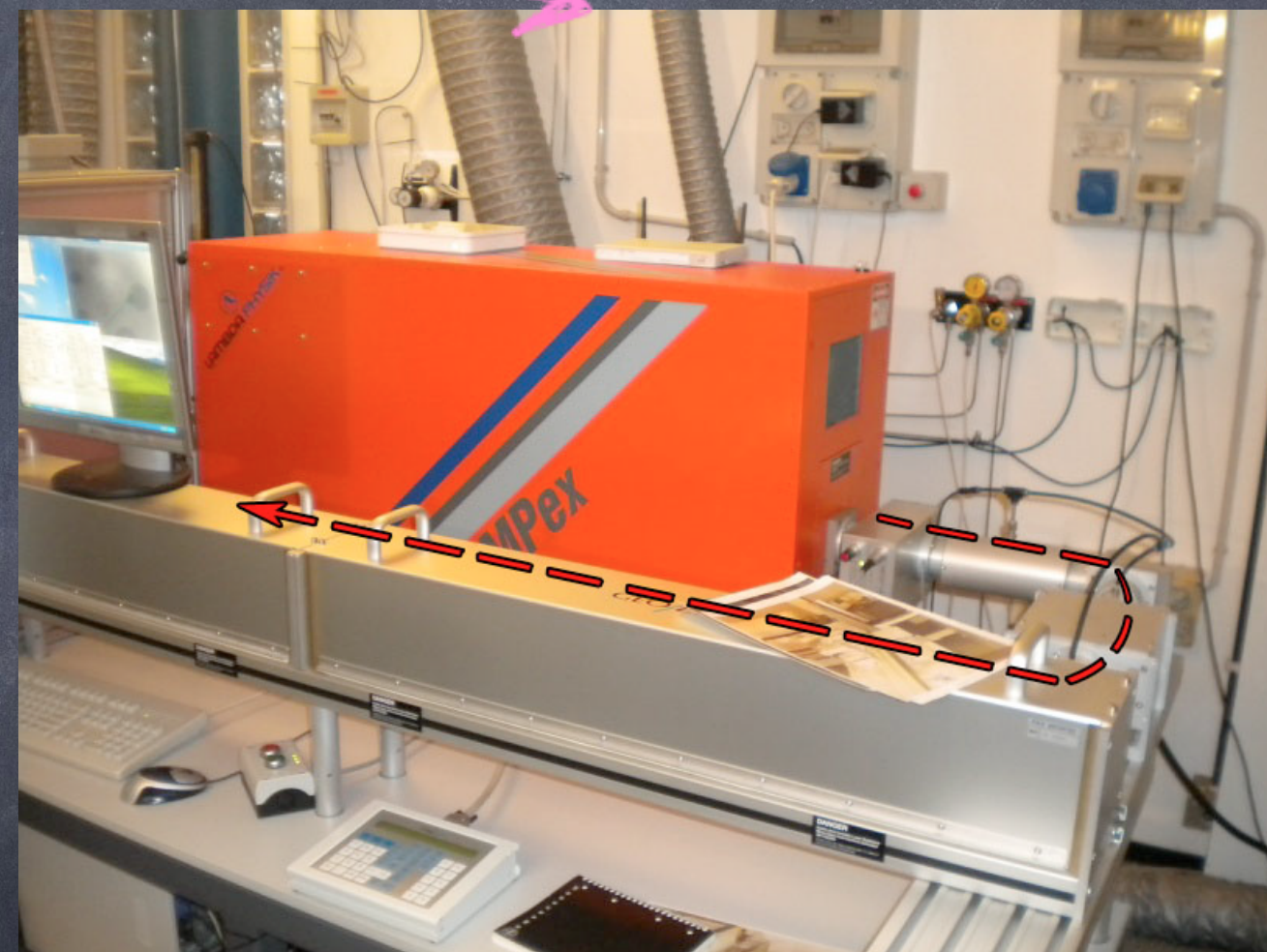
*Consiglio Nazionale delle Ricerche  
Istituto di Geoscienze e Georisorse – Pavia*



# - Outline

- - Principles of LA-ICP-MS technique
- - Trace Element analysis
- - U-Th-Pb geochronology

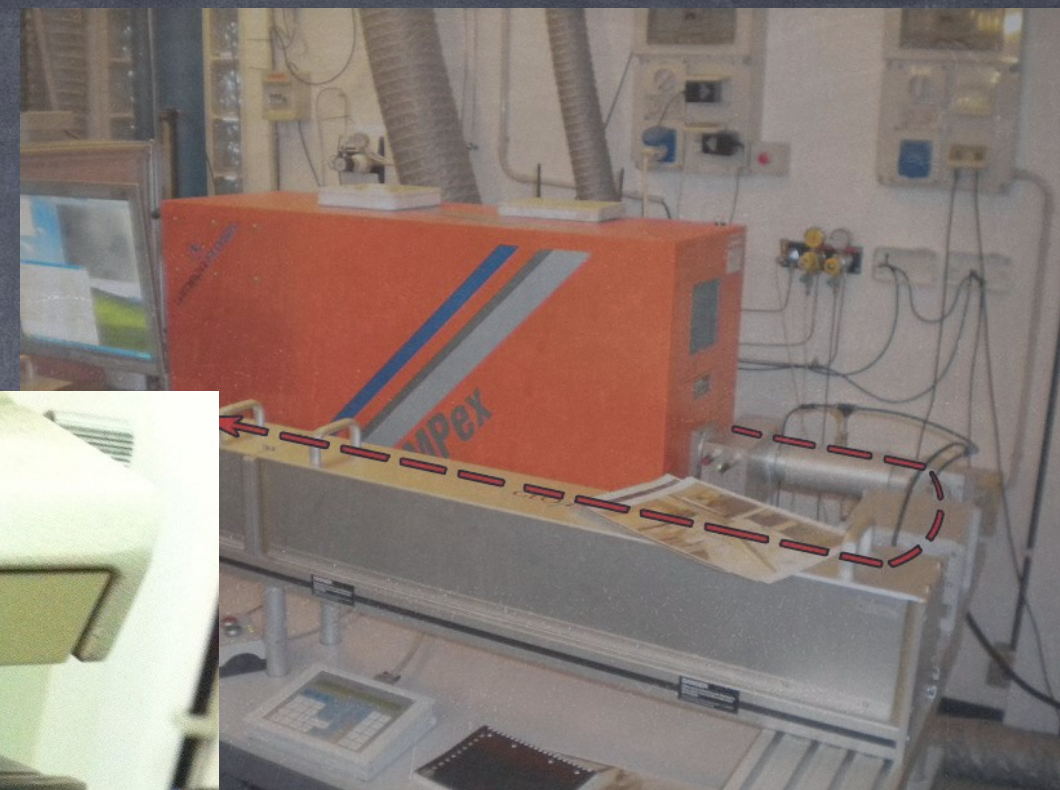
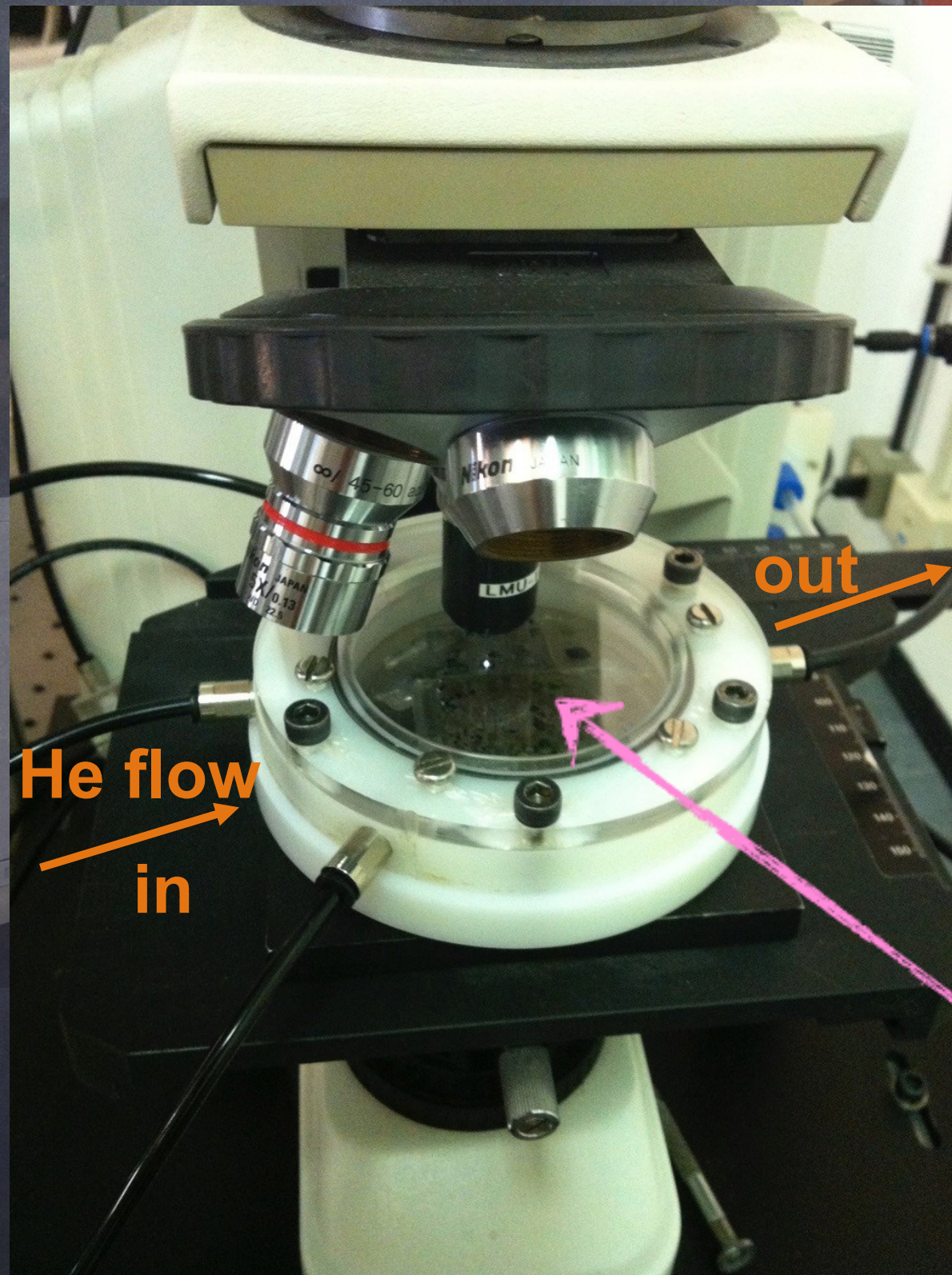
# LA + ICP + MS



8900 QQQ (Agilent)

Excimer Laser 193nm ArF (GeoLas 102, Microlas)

# LA + ICP + MS



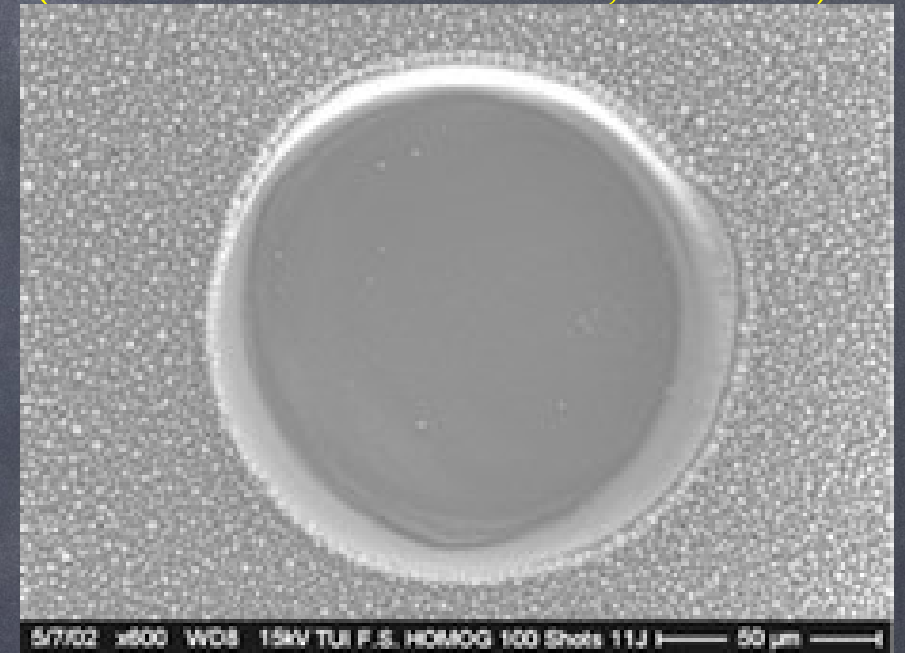
Nd:YAG laser

Ablation cell

# LA-ICP-MS approaches

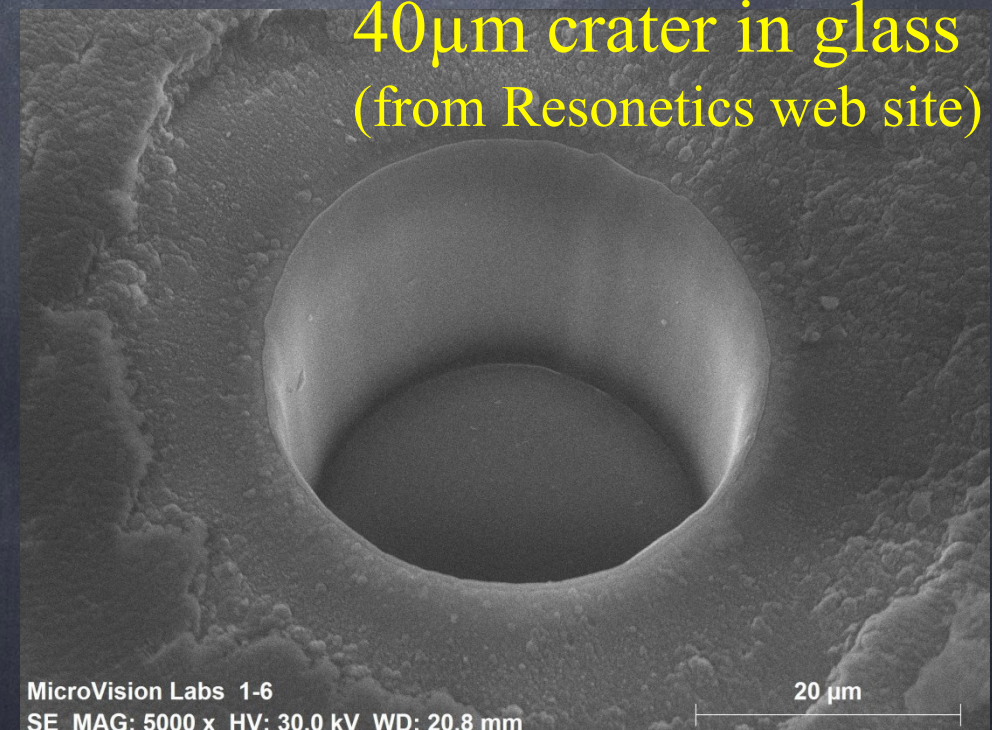
“In situ”: isotopic analysis of  
small cylindrical volumes  
(5-100  $\mu\text{m}$  diameter)

100  $\mu\text{m}$  crater in fused silica  
(from Coherent web site, 193 nm)

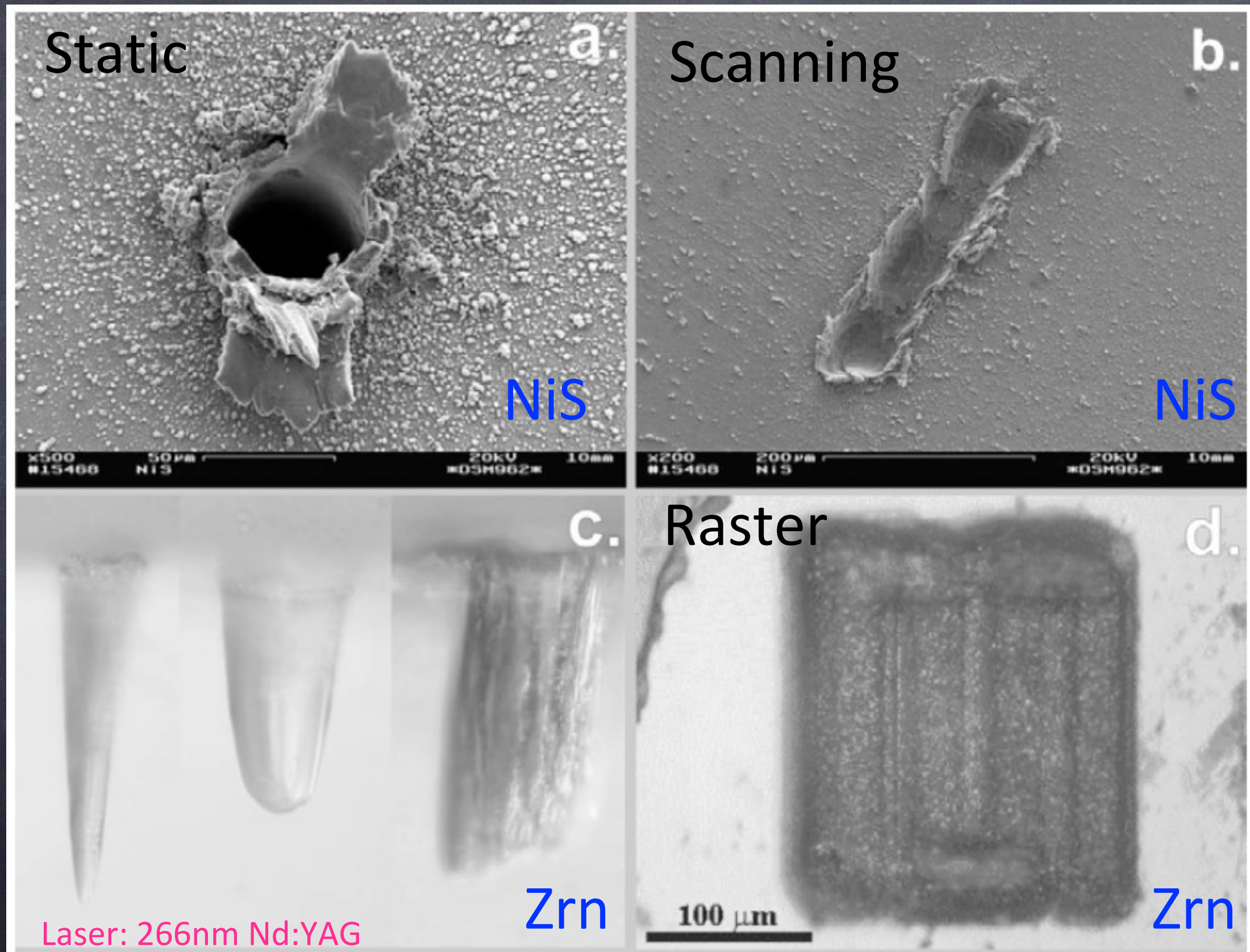


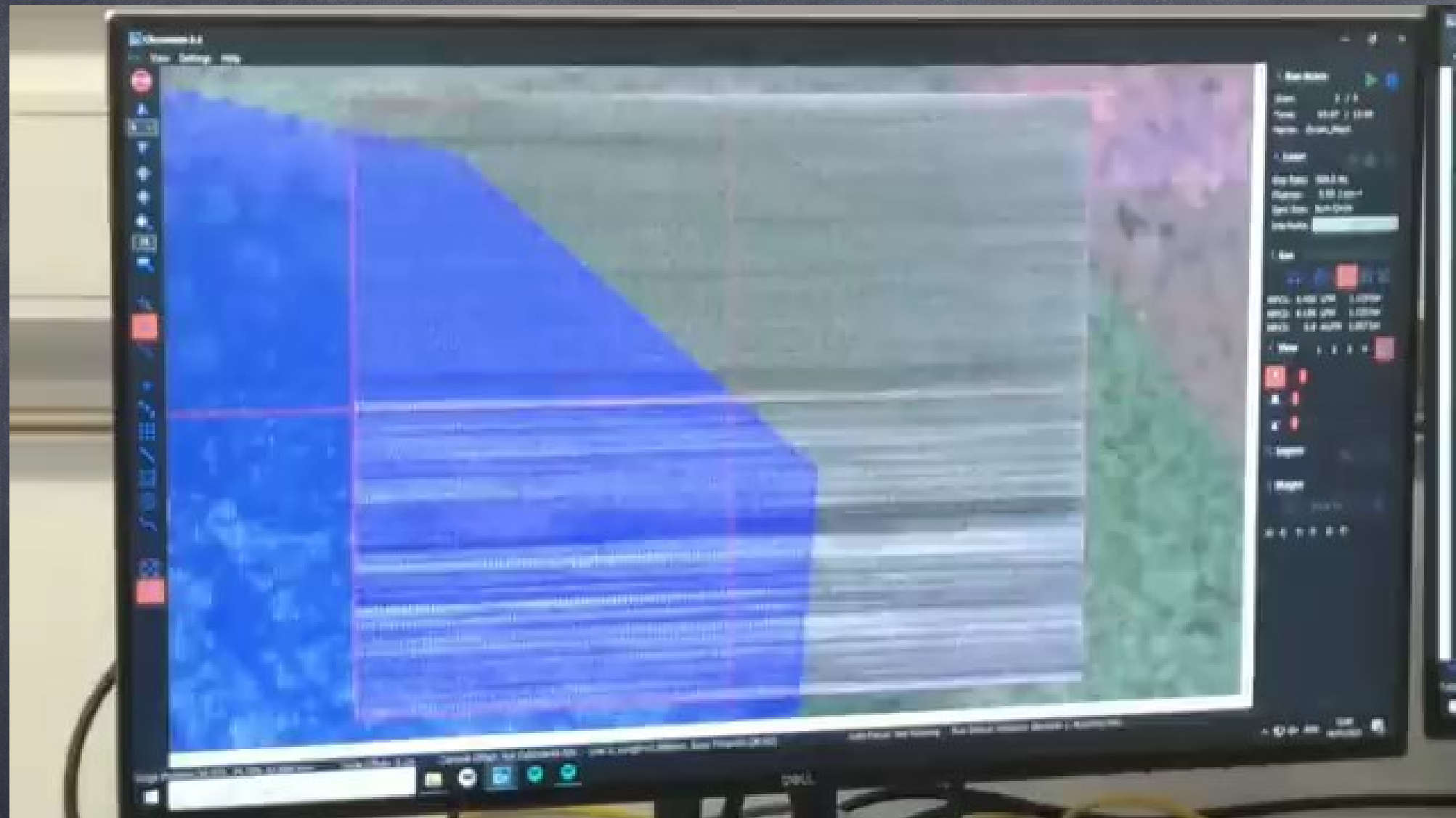
Mineral/solid material  
**separates** or  
directly on ordinary **thin**  
**sections**

40  $\mu\text{m}$  crater in glass  
(from Resonetics web site)



# Ablation protocols





**Analisi raster di zircone con Excimer LASER (193 nm).  
Spot 5  $\mu\text{m}$ , 200 KHz**

**(Irydia, Teledyne, Spot da 1  $\mu\text{m}$ , 1 MHz)**

# Q-and-QQQ-ICP-MS

Agilent Series 8900 QQQ



Quadrupolo (Q)

Reaction/Collision Cell (R/CC)  
= rimozione interferenze  
isobariche

## ICP-MS a triplo quadrupolo Agilent con MS/MS

Filtro di massa quadrupolare (Q1)

Scarta tutti gli ioni fuori massa prima  
che possano entrare nella cella

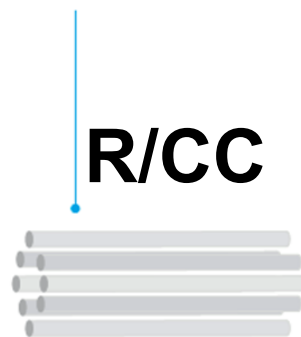


**Q1**

Solo l'analita e le  
interferenze sulla massa  
raggiungono la cella

Cella di reazione

Usa la chimica di reazione per risolvere  
l'analita dalle interferenze sulla massa



**R/CC**

Solo gli ioni dell'analita (e gli  
ioni fuori massa/ioni prodotto)  
escono dalla cella

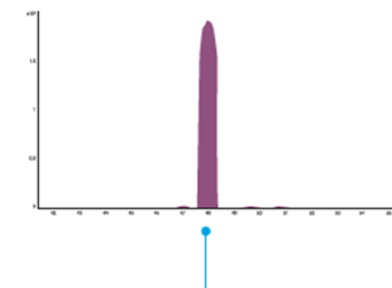
Filtro di massa quadrupolare (Q2)

Scarta gli ioni fuori massa e lascia passare  
al rivelatore l'analita senza interferenze

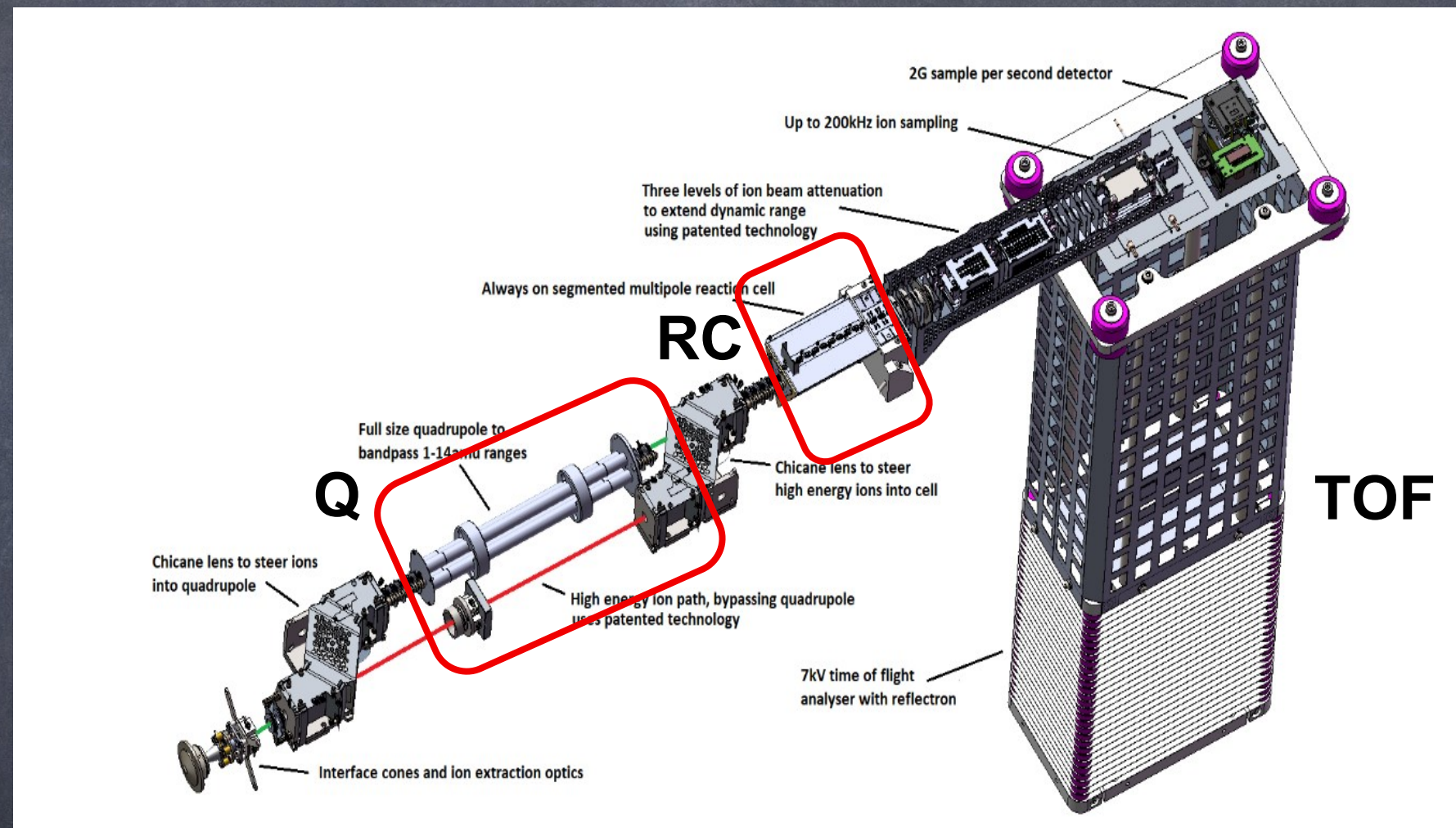


**Q2**

Solo gli ioni dell'analita target o gli ioni  
prodotto contribuiscono al segnale misurato,  
cosicché i risultati sono accurati e coerenti



# ICP-MS-TOF



- scansione velocissima dello spettro in massa a partire da una soglia con acquisizione di tutte le masse

# ICP-MS a Settore Magnetico

## ISTITUTO DI GEOSCIENZE E GEORISORSE (IGG) - PISA

### MULTI COLLECTOR ICP MASS SPECTROMETRY LABORATORY

SAMPLE PREPARATION

SAMPLE WEIGHTING

SAMPLE DIGESTION

CHEMICAL SEPARATION

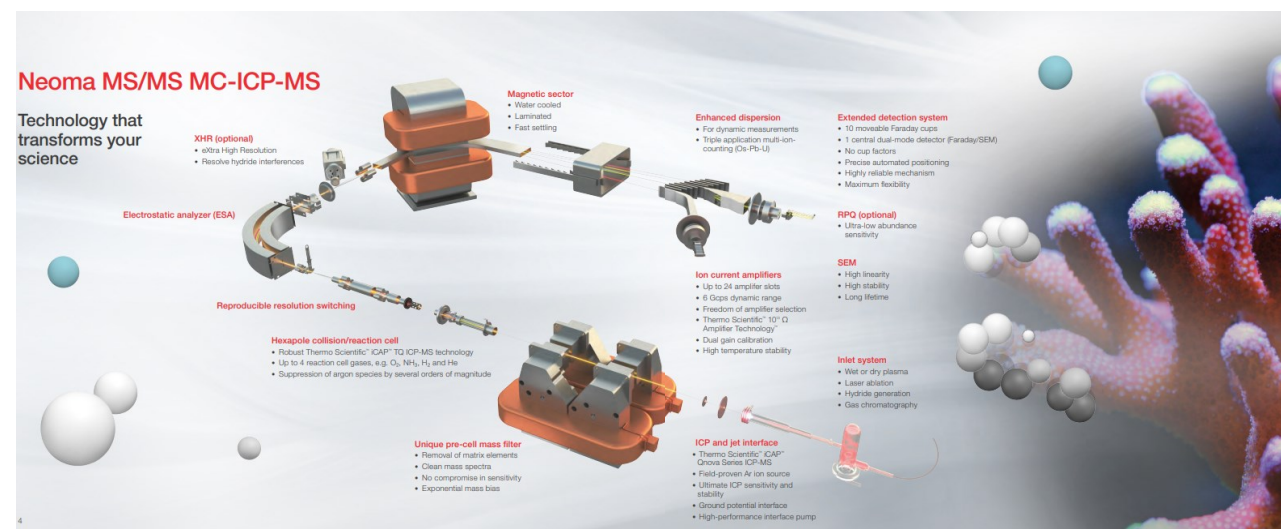
MEASUREMENTS



Thermo Scientific NEOMA  
MS/MS MC-ICP-MS

IN SITU ANALYSES

INTERFERENCE SEPARATION



11 Faraday Cup mobile detector array

10<sup>13</sup> Ω Amplifier Technology

Highest sensitivity and stability

Superior Vacuum System

Extra High Resolution (XHR) increasing the resolving power

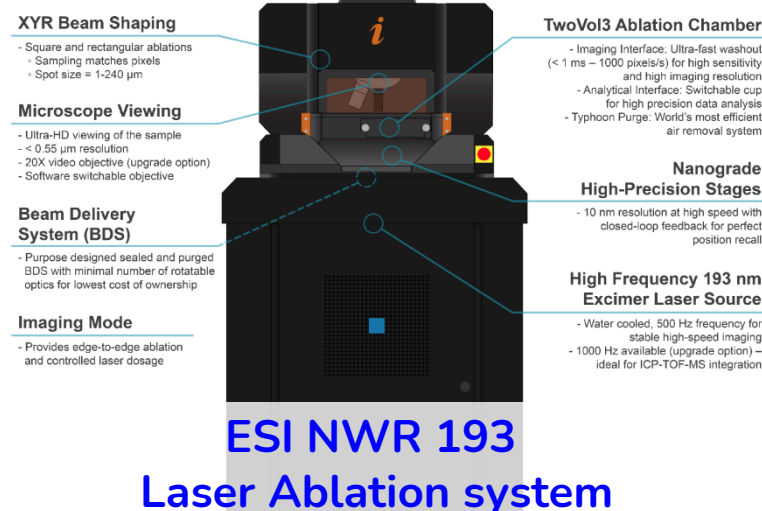
Retarding Potential Quadrupole (RPQ) improving abundance sensitivity

# ICP-MC-MS

## ISTITUTO DI GEOSCIENZE E GEORISORSE (IGG)

NEPTUNE-NEOMA LABORATORY

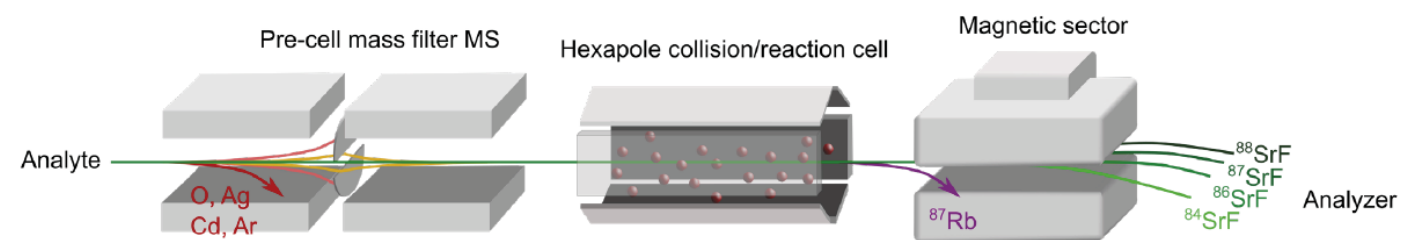
### IN SITU ANALYSES



MINIMAL SAMPLE PREPARATION

SAMPLE CONTAMINATION REDUCTION

### MS/MS TECHNOLOGY



**FILTER**

**REACTION**

**SEPARATION**

REMOVAL OF SAMPLE MATRIX

EFFECTIVE INTERFERENCE REMOVAL

ROBUST ISOTOPE SEPARATION AND DETECTION

NON-ISOBARIC INTERFERENCE REMOVAL

DIFFERENT REACTIVE GASES

SIMULTANEOUS MEASUREMENT OF ALL ISOTOPES OF INTEREST

MAGNETIC SECTOR TECHNOLOGY

LOW TO HIGH RESOLUTION MODE



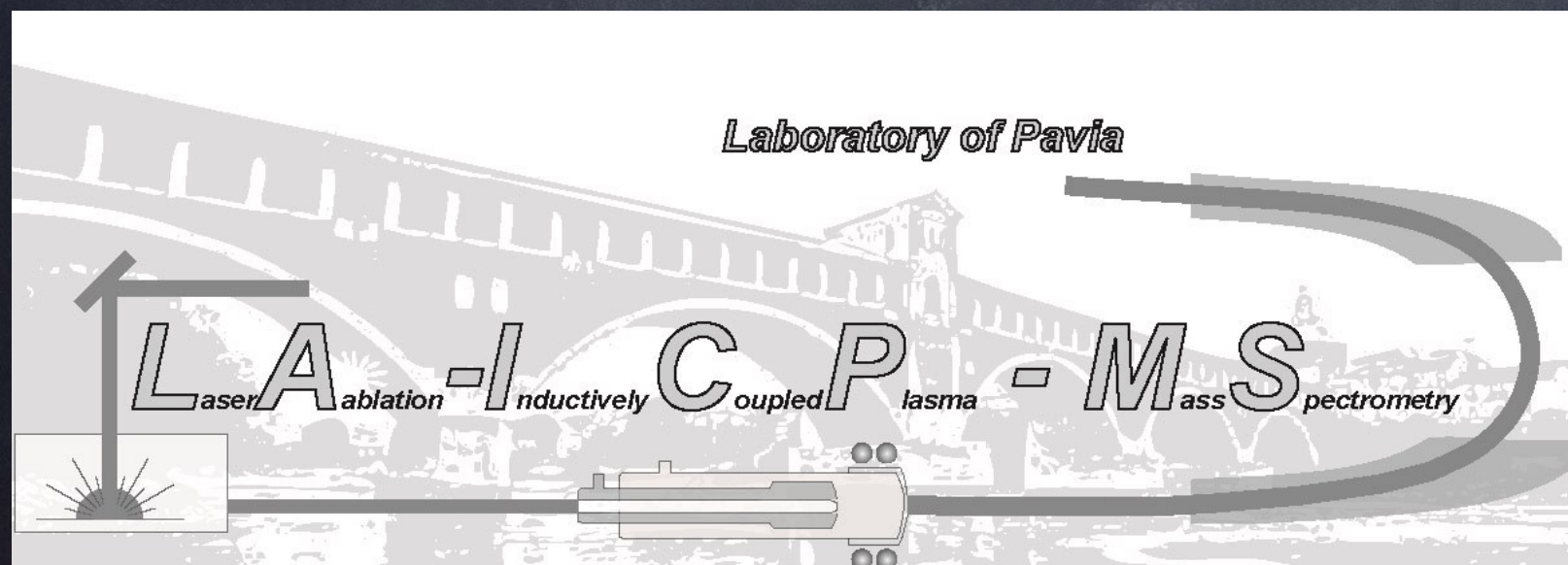
# LA-ICP-MS analysis:

— what we can do!



Trace element  
( $<1000$  ppm/ 0.01 wt%)

U-(Th)-Pb  
Geochronology



# LA-ICP-MS Trace element

|   | 1                 | 2                 | 3                       | 4                 | 5                 | 6                | 7                  | 8                  | 9                 | 10                | 11                | 12                | 13                | 14                 | 15                | 16                | 17              | 18 |   |
|---|-------------------|-------------------|-------------------------|-------------------|-------------------|------------------|--------------------|--------------------|-------------------|-------------------|-------------------|-------------------|-------------------|--------------------|-------------------|-------------------|-----------------|----|---|
| 1 | H                 |                   |                         |                   |                   |                  |                    |                    |                   |                   |                   |                   |                   |                    |                   |                   |                 | He | 1 |
| 2 | Li<br>7<br>40.2   | Be<br>9<br>37.5   |                         |                   |                   |                  |                    |                    |                   |                   |                   |                   |                   |                    |                   |                   |                 |    | 2 |
| 3 | Na<br>23<br>3.23* | Mg<br>24<br>3.56* |                         |                   |                   |                  |                    |                    |                   |                   |                   |                   |                   |                    |                   |                   |                 |    | 3 |
| 4 | K<br>39<br>14900  | Ca<br>43<br>11.9* | Sc<br>45<br>39.9        | Ti<br>49<br>39.4  | V<br>51<br>38.8   | Cr<br>53<br>39.0 | Mn<br>55<br>38.7   | Fe<br>57<br>12.4*  | Co<br>59<br>35.5  | Ni<br>60<br>38.8  | Cu<br>65<br>37.8  | Zn<br>66<br>39.1  | Ga<br>71<br>36.7  | Ge<br>72<br>36.1   | As<br>75<br>35.7  | Se<br>77<br>15.2  | Br<br>79<br>nrv | Kr | 4 |
| 5 | Rb<br>85<br>31.4  | Sr<br>88<br>78.4  | Y<br>89<br>38.3         | Zr<br>90<br>37.9  | Nb<br>93<br>38.9  | Mo<br>95<br>37.4 | Tc<br>99<br>0.0011 | Ru<br>99<br>0.0011 | Rh<br>103<br>0.91 | Pd<br>105<br>1.05 | Ag<br>107<br>22   | Cd<br>111<br>28.1 | In<br>115<br>38.9 | Sn<br>118<br>38.6  | Sb<br>121<br>34.7 | Te<br>125<br>30.9 | I<br>127<br>nrv | Xe | 5 |
| 6 | Cs<br>133<br>42.7 | Ba<br>137<br>39.3 | L<br>177<br>Lanthanides | Hf<br>177<br>36.7 | Ta<br>181<br>37.6 | W<br>182<br>38   | Re<br>185<br>6.63  | Os<br>189<br>nrv   | Ir<br>193<br>nrv  | Pt<br>195<br>2.51 | Au<br>197<br>4.77 | Hg<br>201<br>nrv  | Tl<br>205<br>14.9 | Pb<br>207<br>38.57 | Bi<br>209<br>30.2 | Po                | At              | Rn | 6 |
| 7 | Fr                | Ra                | A<br>Actinides          | Rf                | Db                | Sg               | Bh                 | Hs                 | Mt                |                   |                   |                   |                   |                    |                   |                   |                 |    | 7 |

Routine analysis  
 Interference correction (ic)  
 Potential (ic, sensitivity, few/no ref. values)

**Interference correction (ic)**  
 Oxide, O    Doubly charged, 2+  
 Argide, Ar    Mass interference, 'z'

Pref. mass    major ic  
**Element symbol**  
 NIST 612    ppm or  
 BCR-2G    \* = wt.%  
 [VG2]

e.g. volcanic glass analysis  
*Danger: Isotope choice is sample specific*

PGEs

nrv = no reference value

Minimal inter-element fractionation (within Groups)

Pronounced 'across Period' increase in transport efficiency during spot analysis (except Group 1 elements)

|   |                 |                    |                   |                   |    |                   |                   |                   |                   |                   |                   |                 |                   |                   |                 |
|---|-----------------|--------------------|-------------------|-------------------|----|-------------------|-------------------|-------------------|-------------------|-------------------|-------------------|-----------------|-------------------|-------------------|-----------------|
| L | 139<br>La<br>36 | 140<br>Ce<br>38.4  | 141<br>Pr<br>37.9 | 146<br>Nd<br>35.5 | Pm | 147<br>Sm<br>37.7 | 153<br>Eu<br>35.6 | 157<br>Gd<br>37.3 | 159<br>Tb<br>37.6 | 163<br>Dy<br>35.5 | 165<br>Ho<br>38.3 | 166<br>Er<br>38 | 169<br>Tm<br>36.8 | 172<br>Yb<br>39.2 | 175<br>Lu<br>37 |
| A | Ac              | 232<br>Th<br>37.79 | Pa                | 238<br>U<br>37.38 | Np | Pu                | Am                | Cm                | Bk                | Cf                | Es                | Fm              | Md                | No                | Lr              |

# LA-ICP-MS Trace element

Main  
application

Silicates: Olivine, Amphibole, Pyroxene, Garnet, Mica, Plagioclase, epidote, titanite..

Oxides: chromite, rutile, corundum

Phosphate (e.g. Monazite)

Sulfides (e.g. Pyrite) & Sulfates (e.g. Anhydrite  $\text{CaSO}_4$ )

Petrology

Carbonates: calcite and dolomite

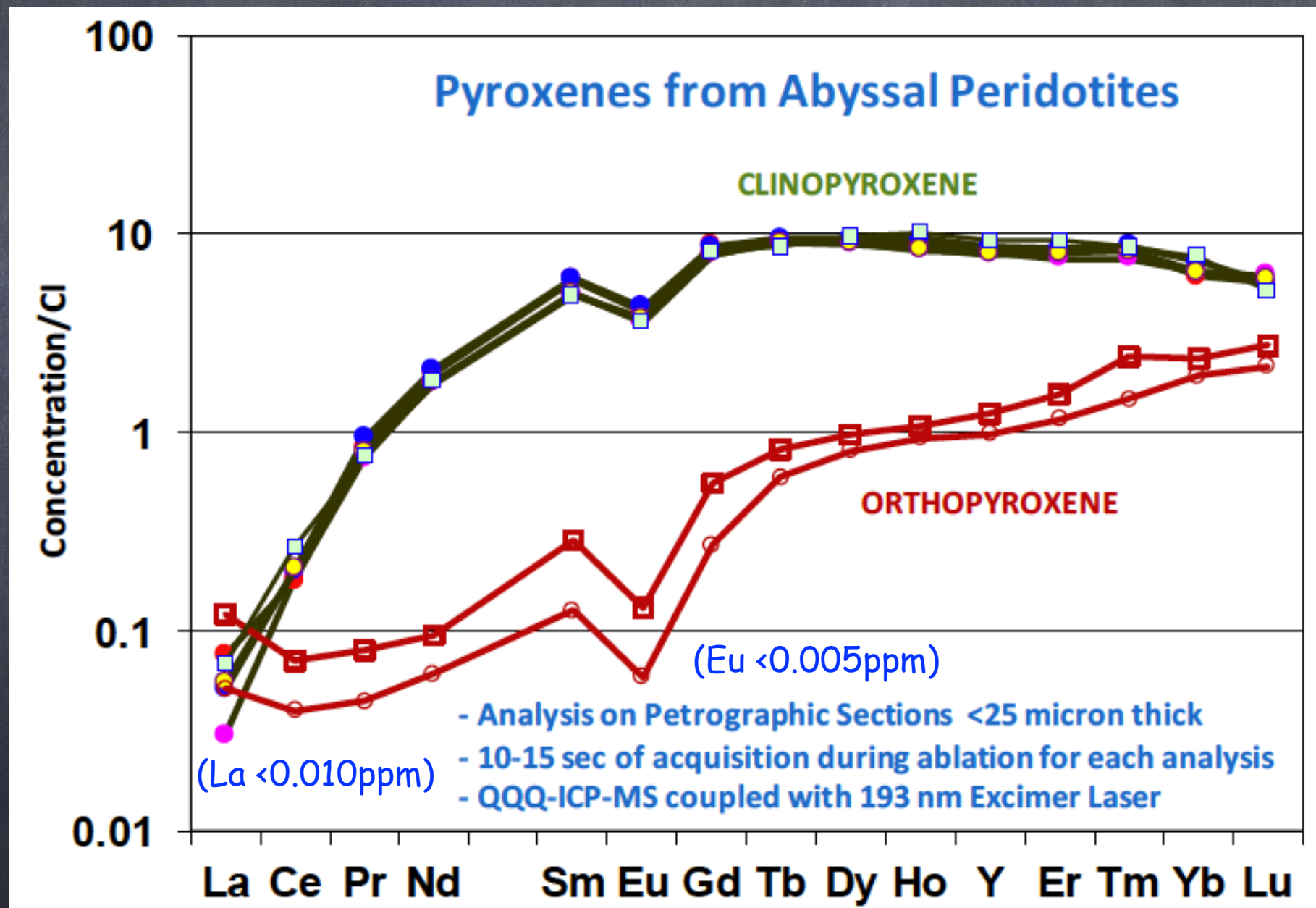
Biology

Glasses

Archaeometry

# LA-ICP-MS Trace element

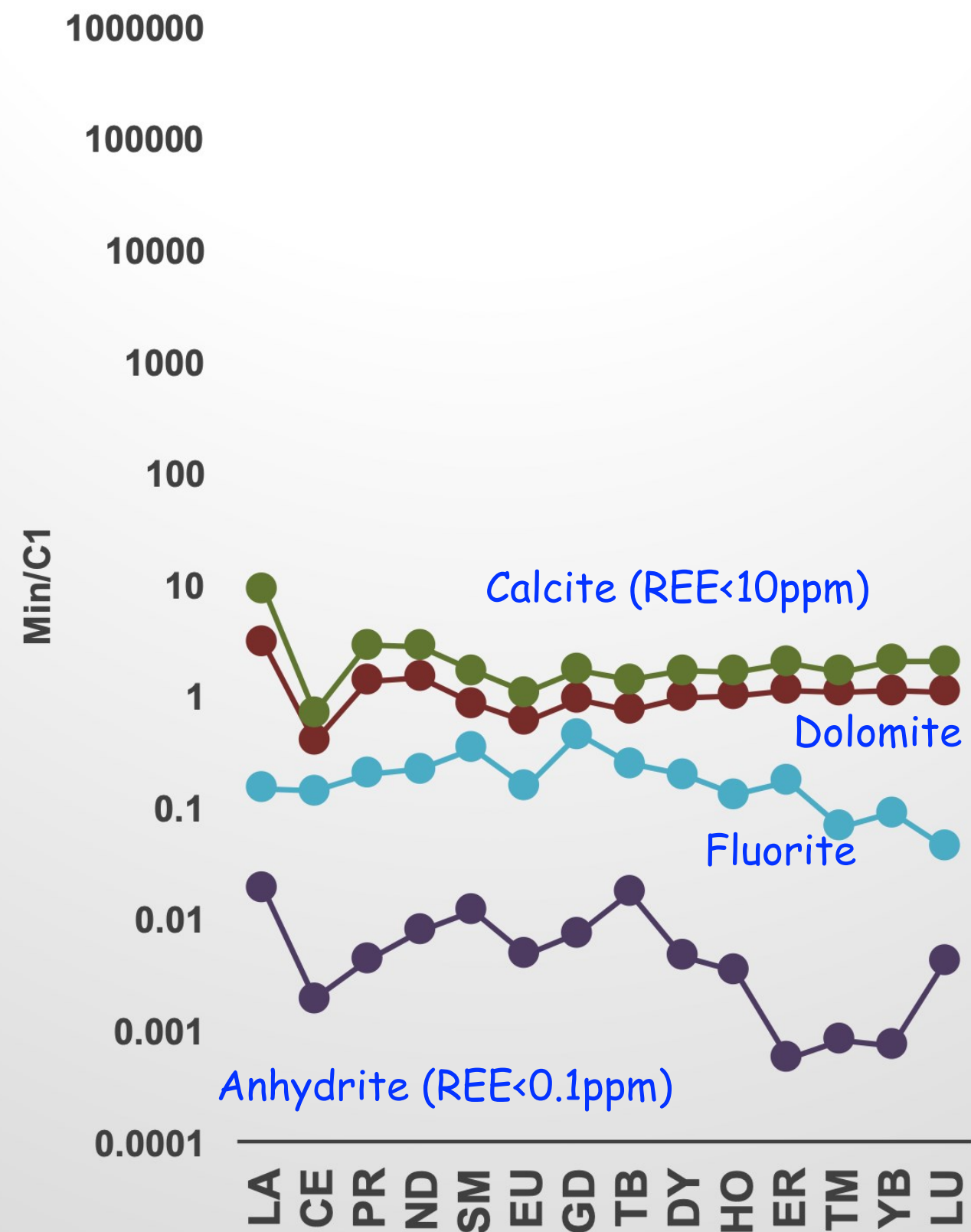
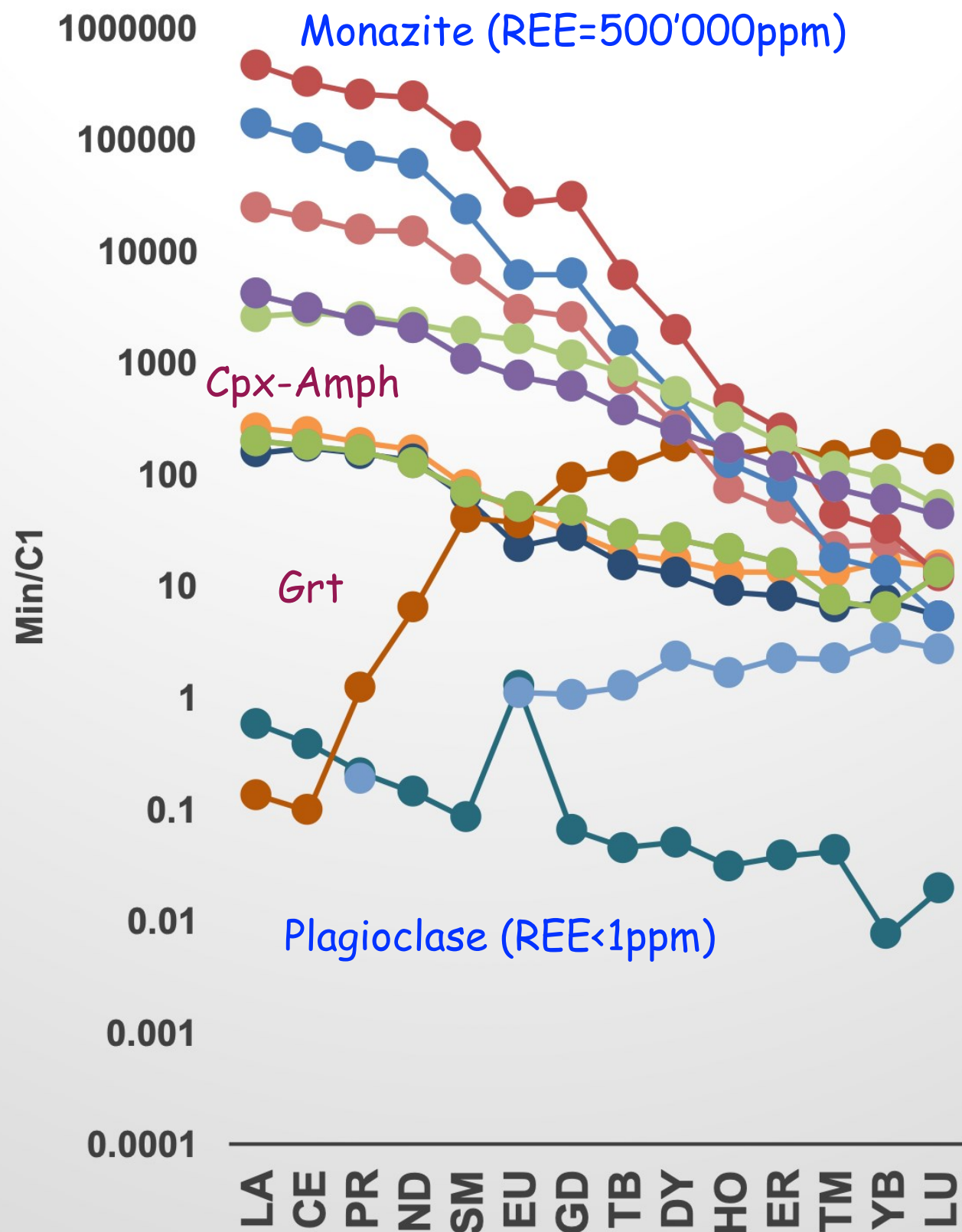
Petrology....



Trace element composition normalised to Chondrite CI (Primitive Meteorites)

# LA-ICP-MS Trace element

Petrology....



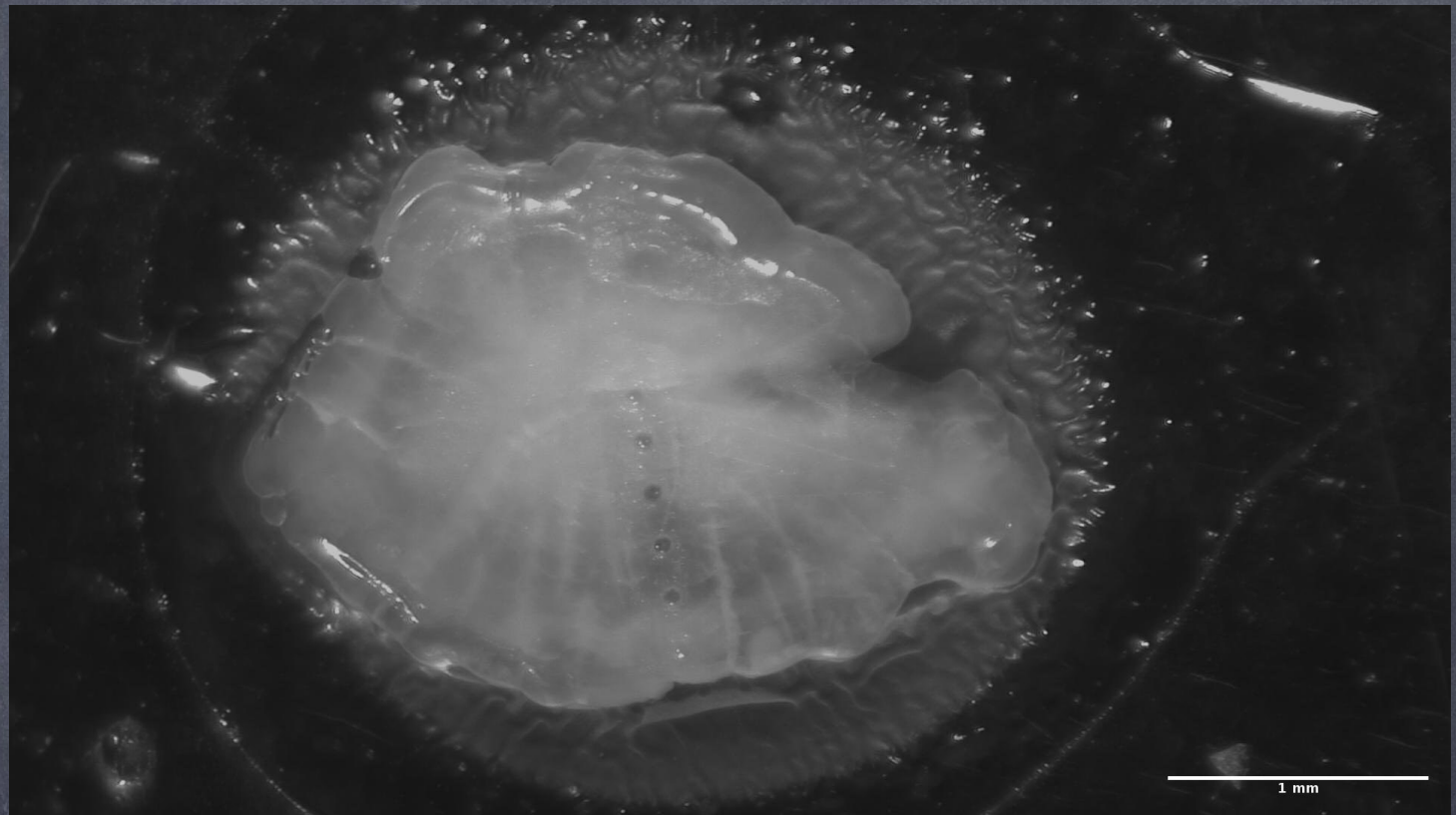
# LA-ICP-MS Trace element

Biology: chemical profiles of carbonate growth



Lithothamnion corallioides  
(Alghe rosse calcaree)

In collaboration with Prof. Basso Daniela Maria (UniMiB)  
and co-authors  
Piazza et al. 2021



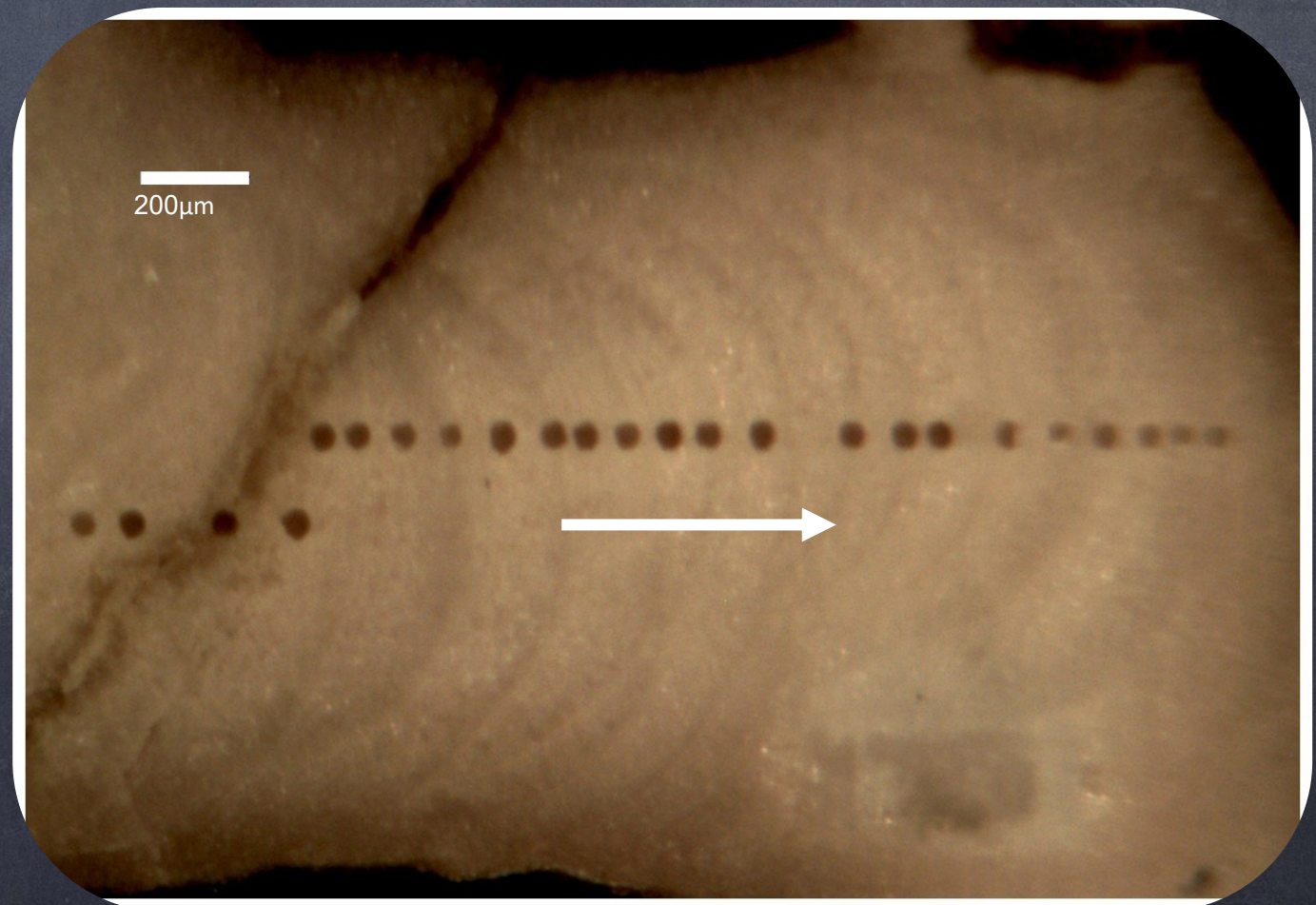
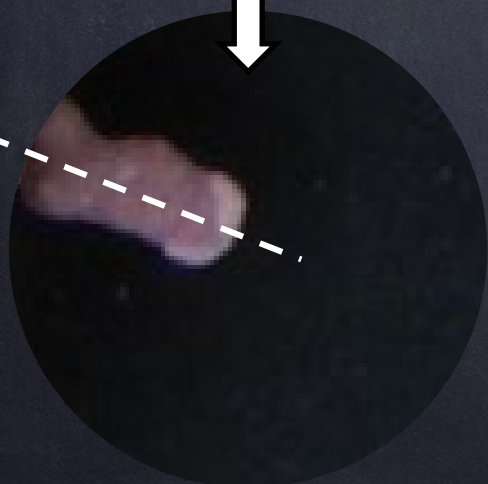
Mullet Otolith  
(Otolite di triglia)

In collaboration with Marco Stagioni (UniBo)

# LA-ICP-MS Trace element

Lithothamnion corallioides  
(Alghe rosse calcaree)

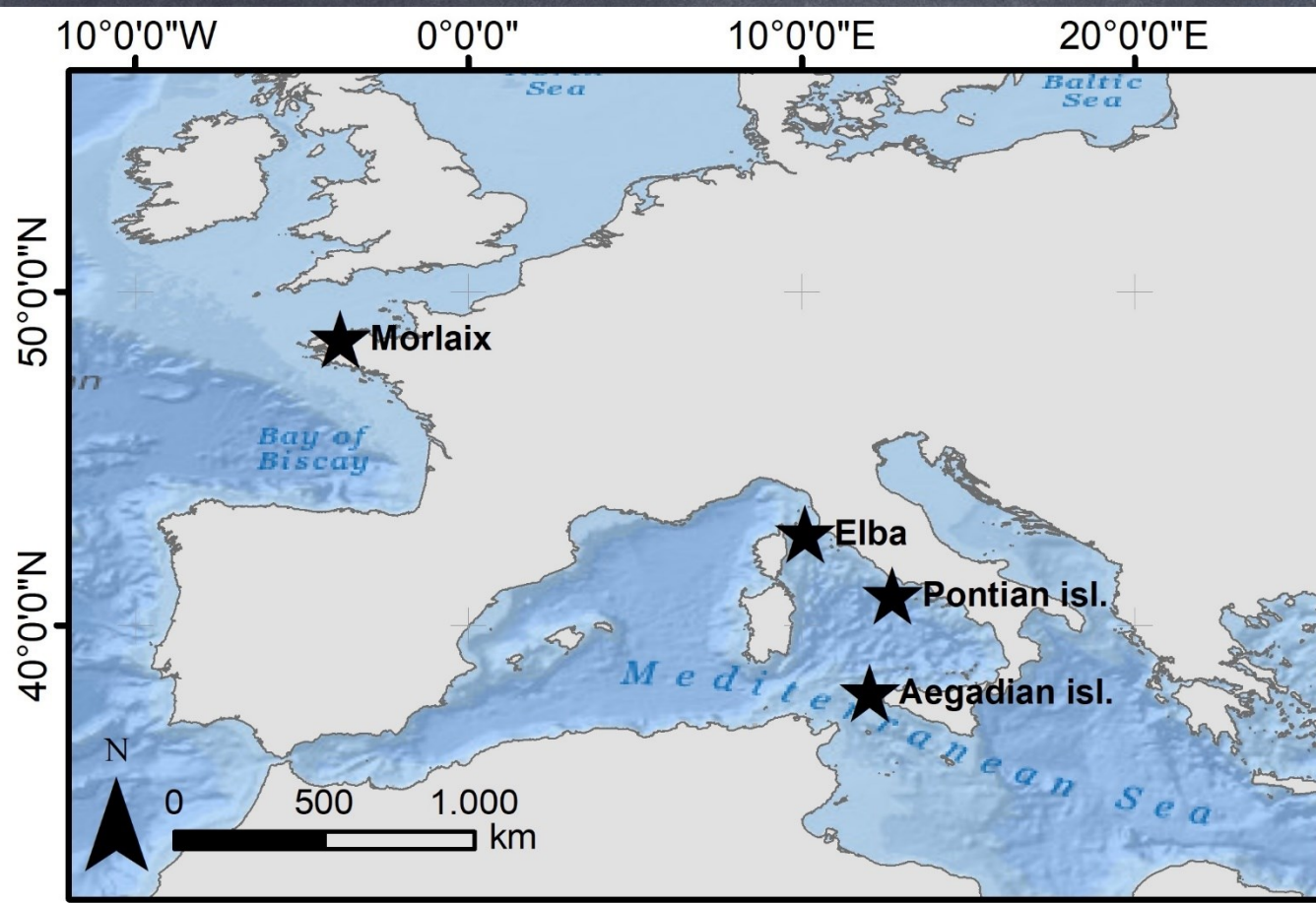
LA-ICP-MS was performed on longitudinal sections of the calcareous red algae *Lithothamnion corallioides*, following the growth direction



In collaboration with Prof. Basso Daniela Maria (UniMiB) and co-authors  
May 2019

# LA-ICP-MS Trace element

Lithothamnion corallioides  
(Alghe rosse calcaree)



$356.0 \mu\text{mol/mol} < \text{B/Ca} < 954.1 \mu\text{mol/mol}$   
 $171.7 \text{ mmol/mol} < \text{Mg/Ca} < 311.2 \text{ mmol/mol}$   
 $53.8 \mu\text{mol/mol} < \text{Li/Ca} < 178.5 \mu\text{mol/mol}$   
 $2.9 \text{ mmol/mol} < \text{Sr/Ca} < 5.1 \text{ mmol/mol}$

| Sampling site | B/Ca<br>( $\mu\text{mol/mol}$ ) |              | Mg/Ca<br>( $\text{mmol/mol}$ ) |             | Li/Ca<br>( $\mu\text{mol/mol}$ ) |             | Sr/Ca<br>( $\text{mmol/mol}$ ) |            | Mg/Li<br>( $\text{mol/mmol}$ ) |            |
|---------------|---------------------------------|--------------|--------------------------------|-------------|----------------------------------|-------------|--------------------------------|------------|--------------------------------|------------|
|               | mean                            | st. dev.     | mean                           | st. dev.    | mean                             | st. dev.    | mean                           | st. dev.   | mean                           | st. dev.   |
| Aegadian Isl. | 610.8                           | 63.9         | 224.9                          | 30.3        | 81.9                             | 18.4        | 3.7                            | 0.4        | 2.8                            | 0.5        |
| Elba          | 757.7                           | 75.5         | 223.4                          | 26.4        | 85.2                             | 14.3        | 3.6                            | 0.4        | 2.7                            | 0.3        |
| Pontian Isl.  | 462.8                           | 49.2         | 216.1                          | 21.9        | 79.6                             | 14.6        | 3.3                            | 0.3        | 2.8                            | 0.4        |
| Morlaix       | 726.9                           | 102.8        | 239.5                          | 41.2        | 113.1                            | 32.9        | 4.1                            | 0.5        | 2.2                            | 0.4        |
| total         | <b>661.9</b>                    | <b>138.9</b> | <b>225.3</b>                   | <b>30.4</b> | <b>89.0</b>                      | <b>23.3</b> | <b>3.7</b>                     | <b>0.5</b> | <b>2.6</b>                     | <b>0.4</b> |

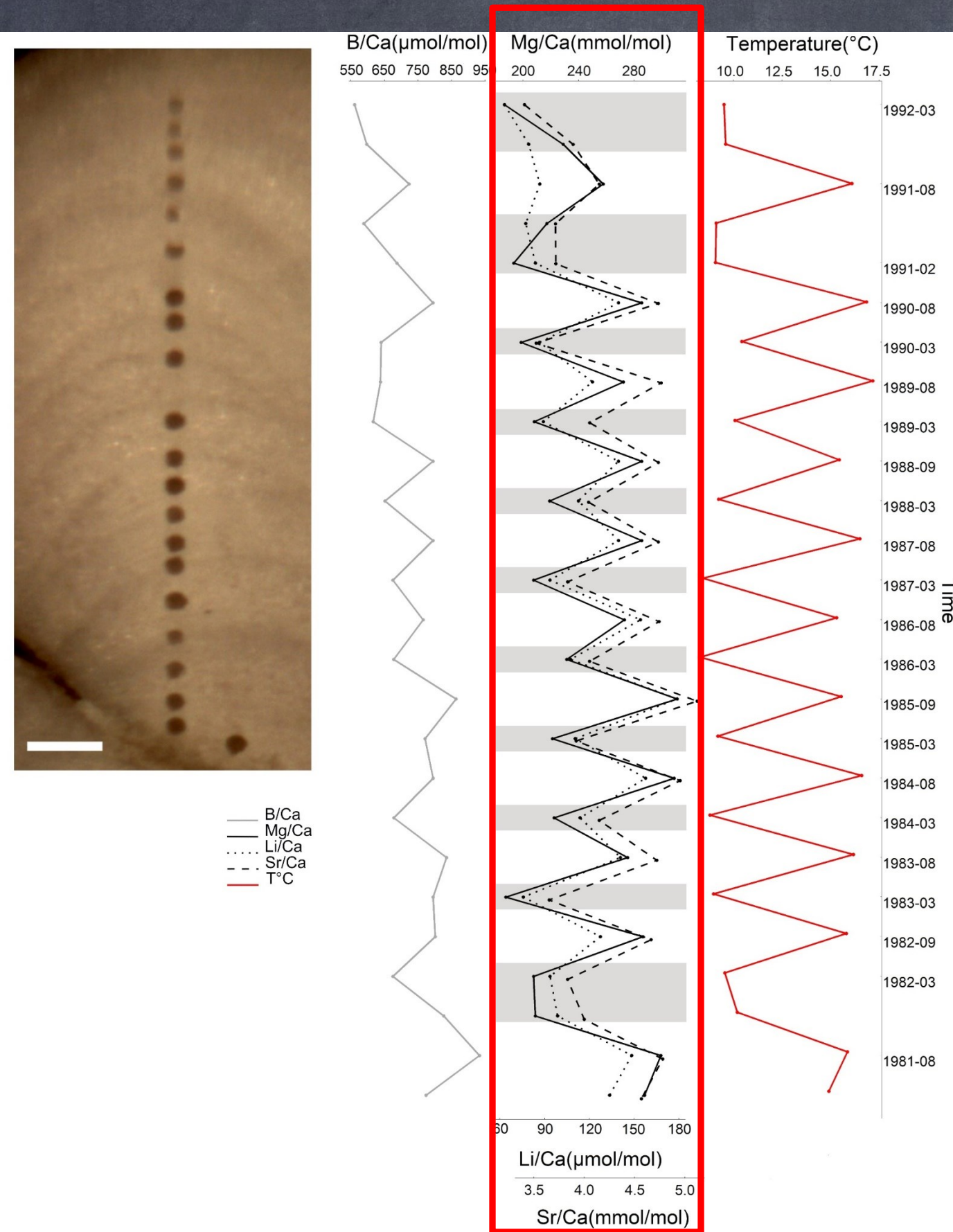
# LA-ICP-MS Trace element

Lithothamnion corallioides  
(Alghe rosse calcaree)

Dark bands correspond to  
winter growth,  
Light bands correspond to  
summer growth.

Mg, Li, Sr/Ca show cyclic  
variations corresponding to the  
seasonal fluctuation in seawater  
temperature.

B/Ca lack of a significant temperature  
influence.

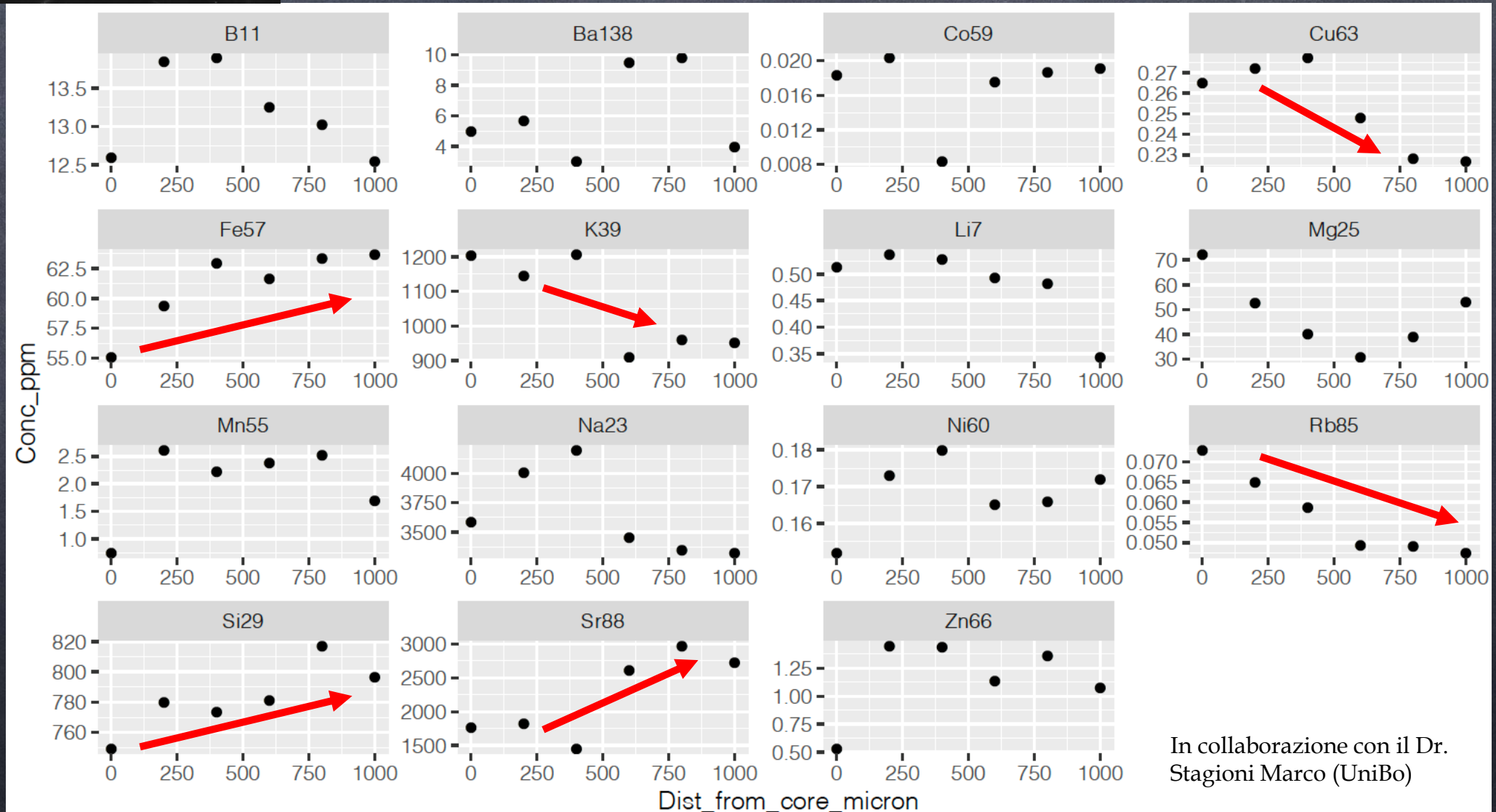
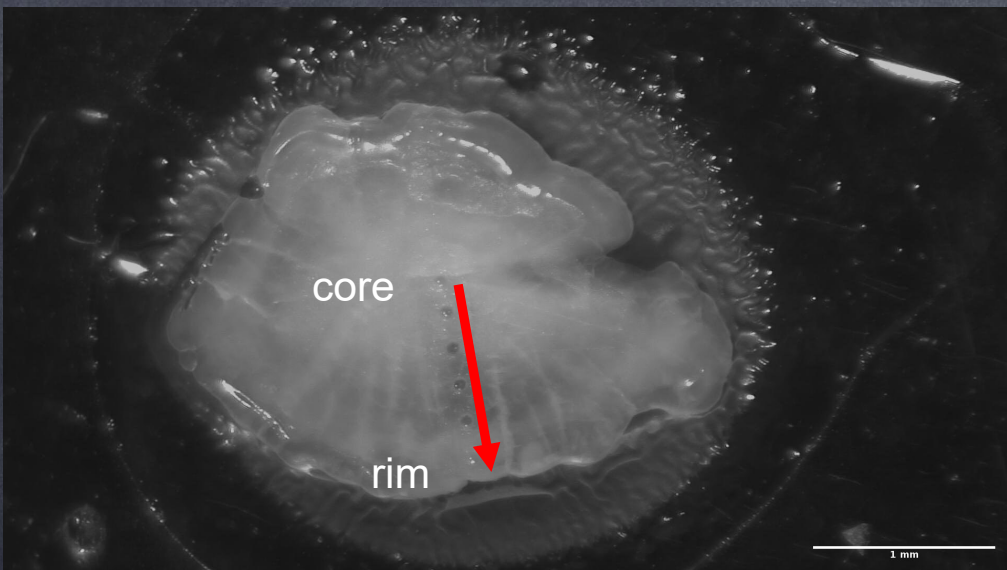


# LA-ICP-MS Trace element

LA-ICP-MS was performed on mullet otholith

Core => larva => pelagic conditions

Rim => 1 year old => deep water conditions



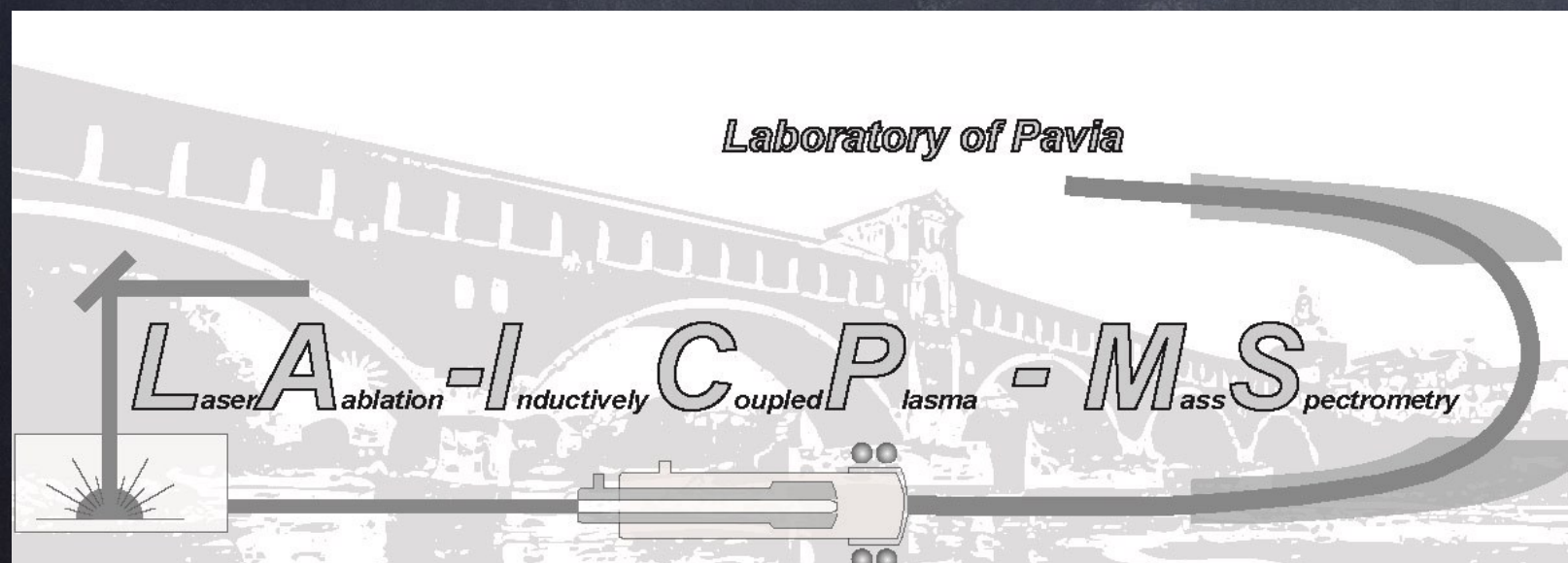
In collaborazione con il Dr.  
Stagioni Marco (UniBo)

# LA-ICP-MS analyses:

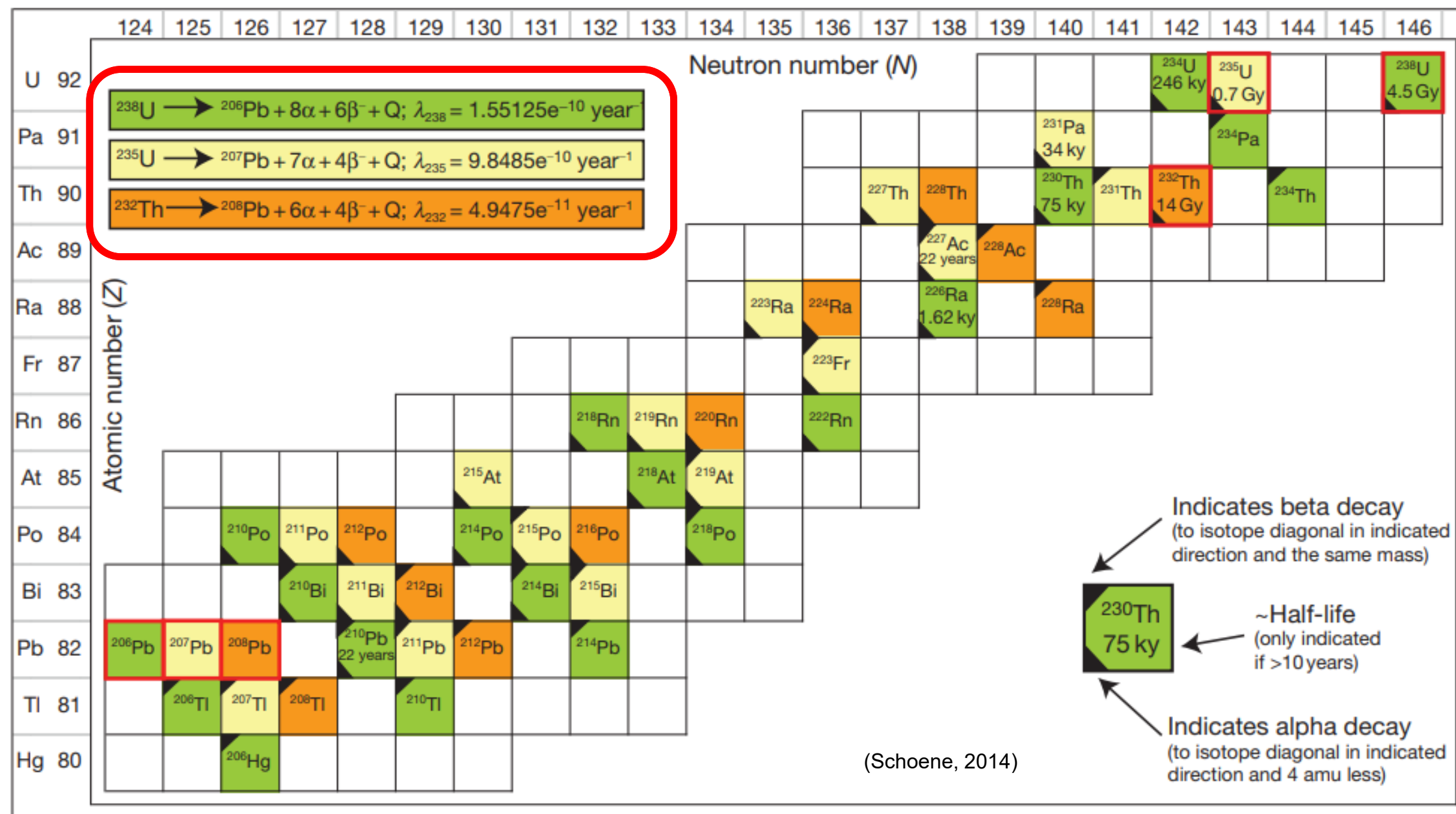
— what we can do!



## U-(Th)-Pb Geochronology

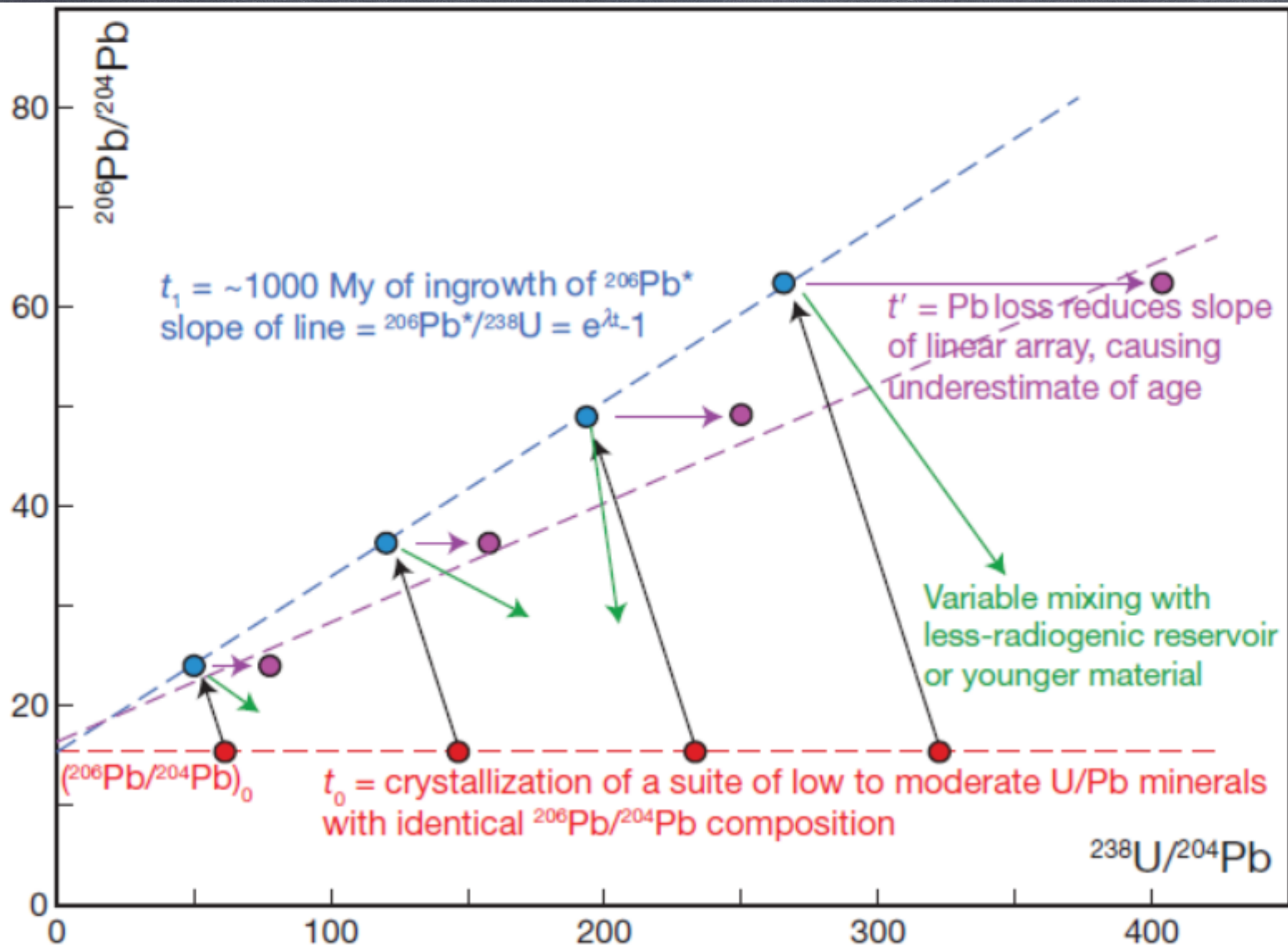


# U-(Th)-Pb Geochronology



**Figure 2** An illustration of the U-Th-Pb decay chains. Each isotope occurring in a given decay chain is color-coded to its parent isotope, which are outlined in red, as are the stable daughter isotopes of Pb. See inset for description of symbols used in each box.  $\alpha$  is an alpha particle,  $\beta$  is a beta particle, and Q is energy released during the decay.

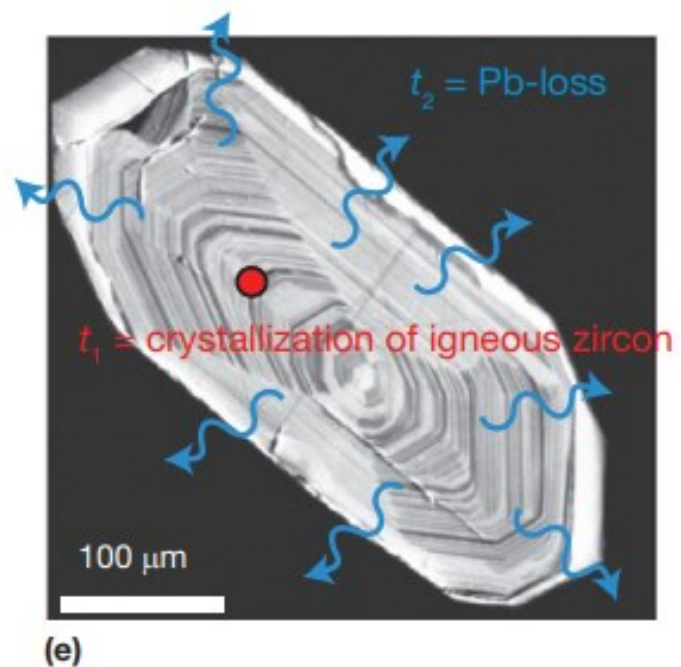
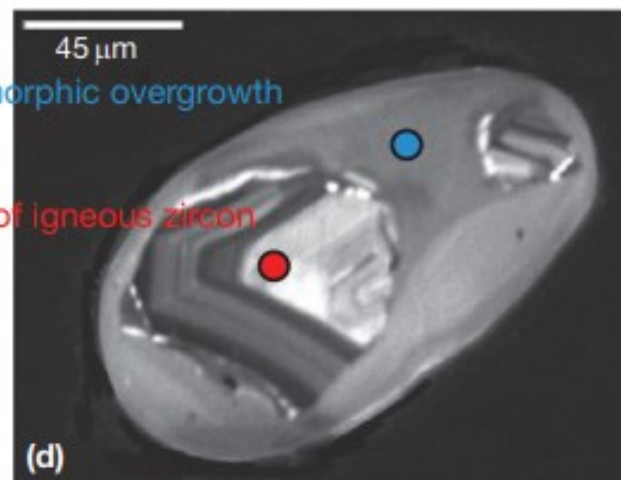
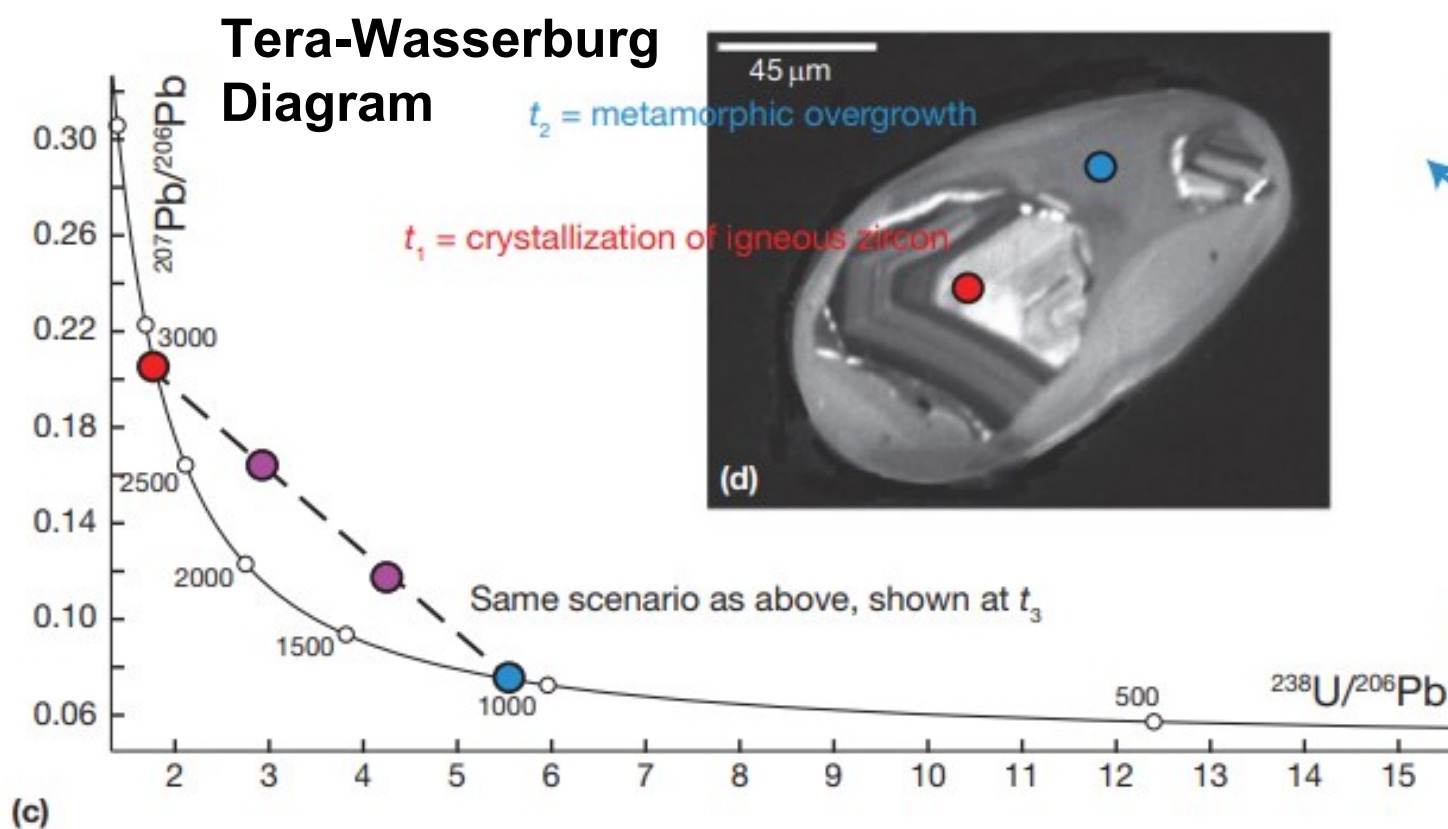
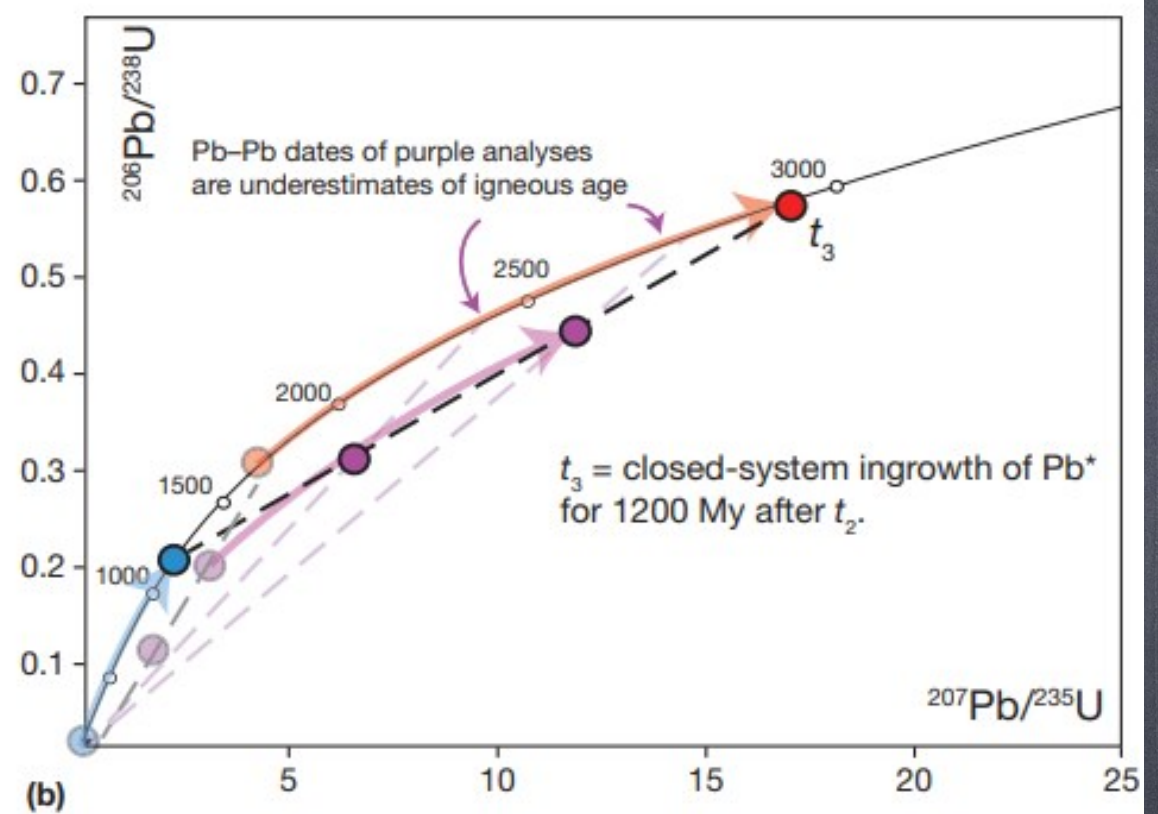
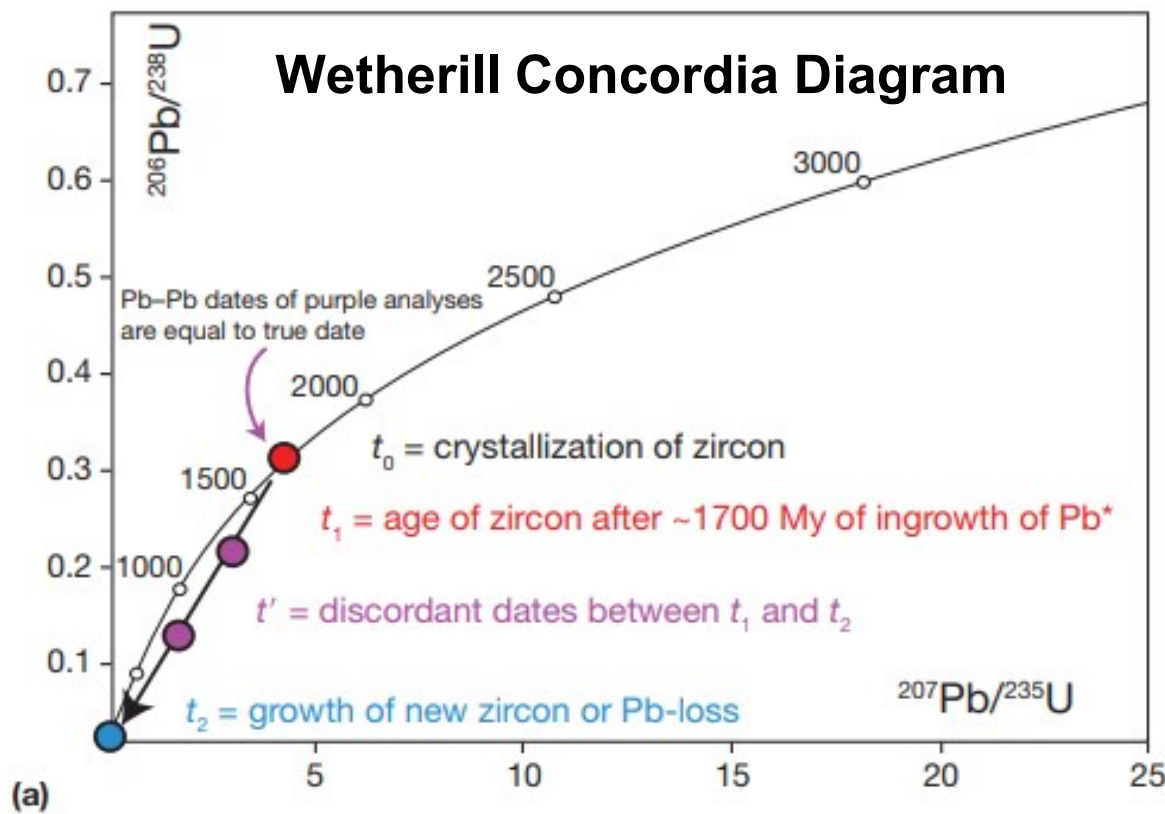
# U-(Th)-Pb Geochronology



(b)

(Schoene, 2014)

# U-(Th)-Pb Geochronology



(Schoene, 2014)

# Common minerals used for U-Pb dating



Zircon:  $\text{ZrSiO}_4$



Monazite:  $(\text{LREE})\text{PO}_4$



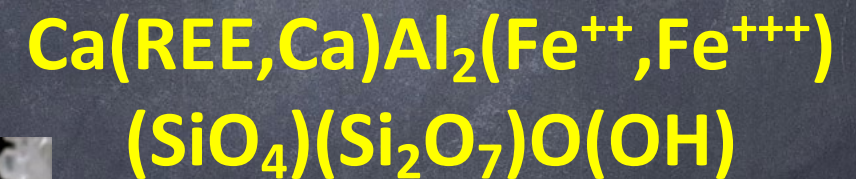
**Rutile** ( $\text{TiO}_2$ )



**Titanite** ( $\text{CaTiSiO}_5$ )



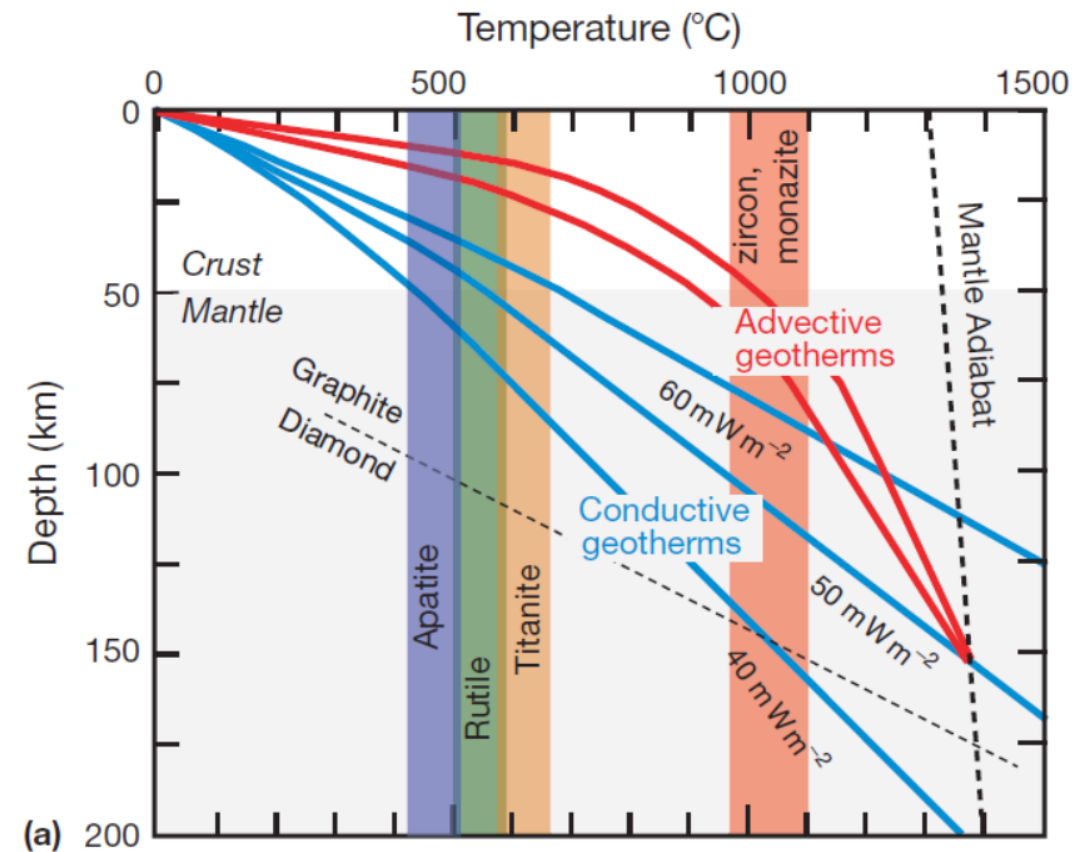
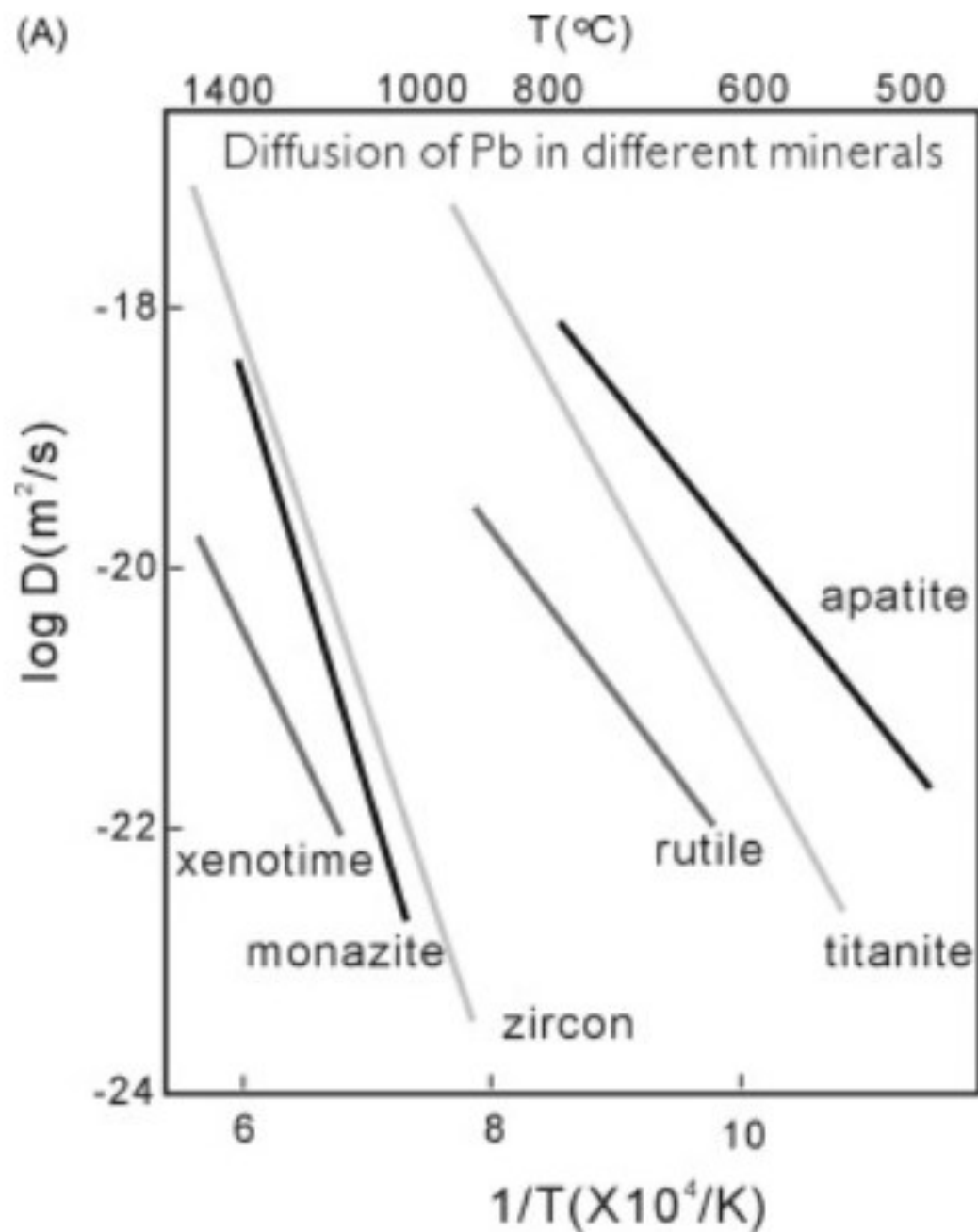
**Allanite**



Th between 0.05 and 3 wt.%  
and U between 10 and 3000  
ppm

prime target for U–Pb and  
Th–Pb geochronology

# U-(Th)-Pb Geochronology



(Schoene, 2014)

Cherniak DJ (2010) Diffusion in accessory minerals: Zircon, Titanite, Apatite, Monazite and Xenotime. *Reviews in Mineralogy & Geochemistry* 72: 827–869.

# .... calcite U-Pb dating

**Calcite**  
( $\text{CaCO}_3$ )

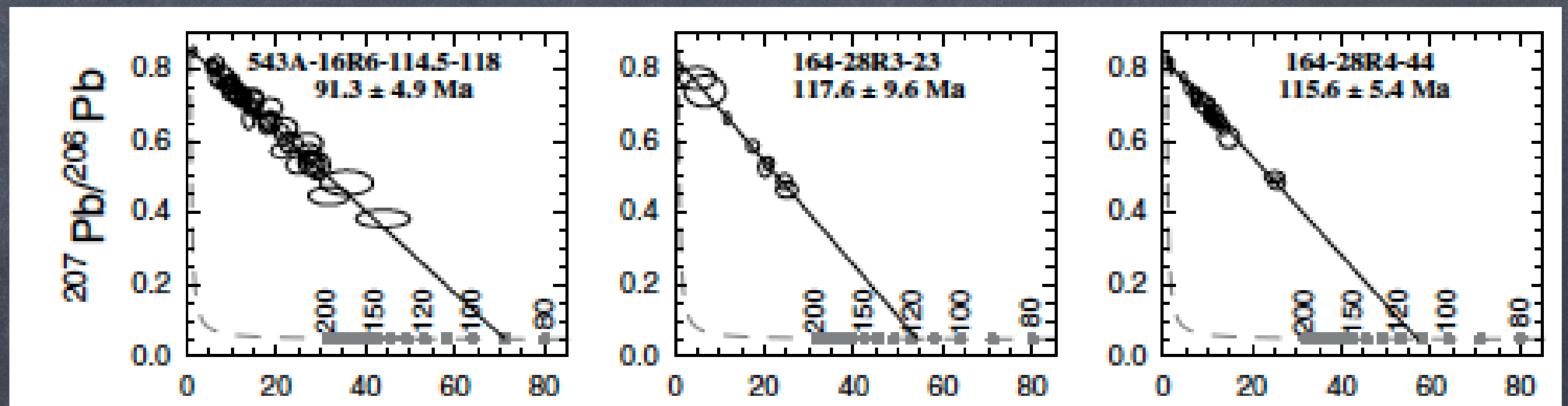
<http://www.geology.neab.net/various/page3.htm>



Early hydrothermal carbon uptake by the upper oceanic crust:  
Insight from in situ U-Pb dating

Geology, 2016

Laurence A. Coogan<sup>1\*</sup>, Randall R. Parrish<sup>2,3</sup>, and Nick M.W. Roberts<sup>3</sup>



**3% 2sigma error**

U ranges from 13 to 2400 ppm!!!!

**14% 2sigma error**

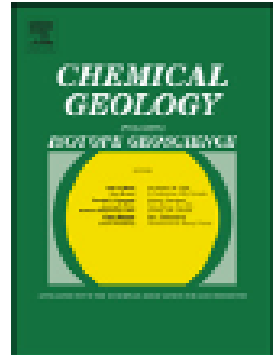
# .... calcite U-Pb dating



Contents lists available at ScienceDirect

Chemical Geology

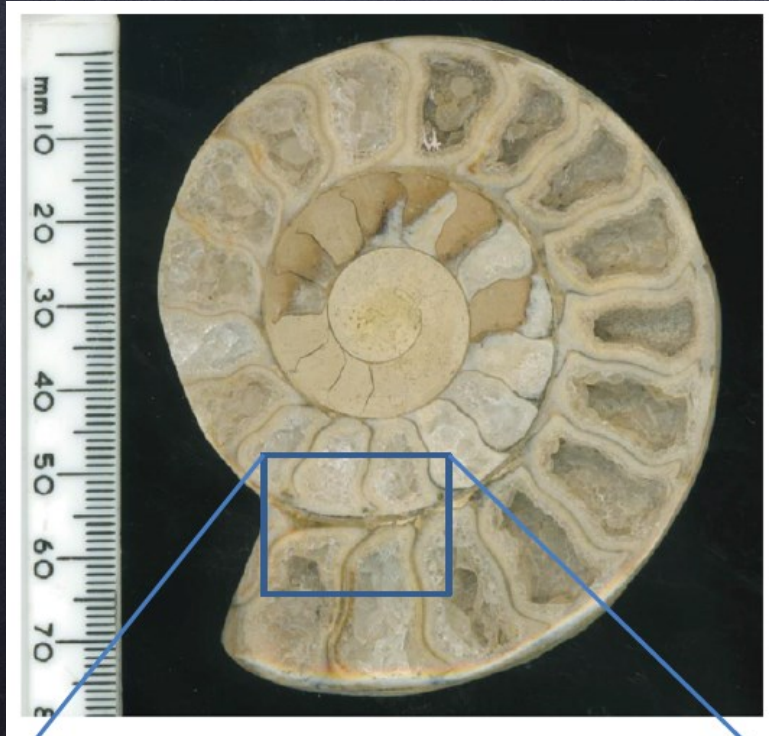
journal homepage: [www.elsevier.com/locate/chemgeo](http://www.elsevier.com/locate/chemgeo)



## U-Pb dating of cements in Mesozoic ammonites

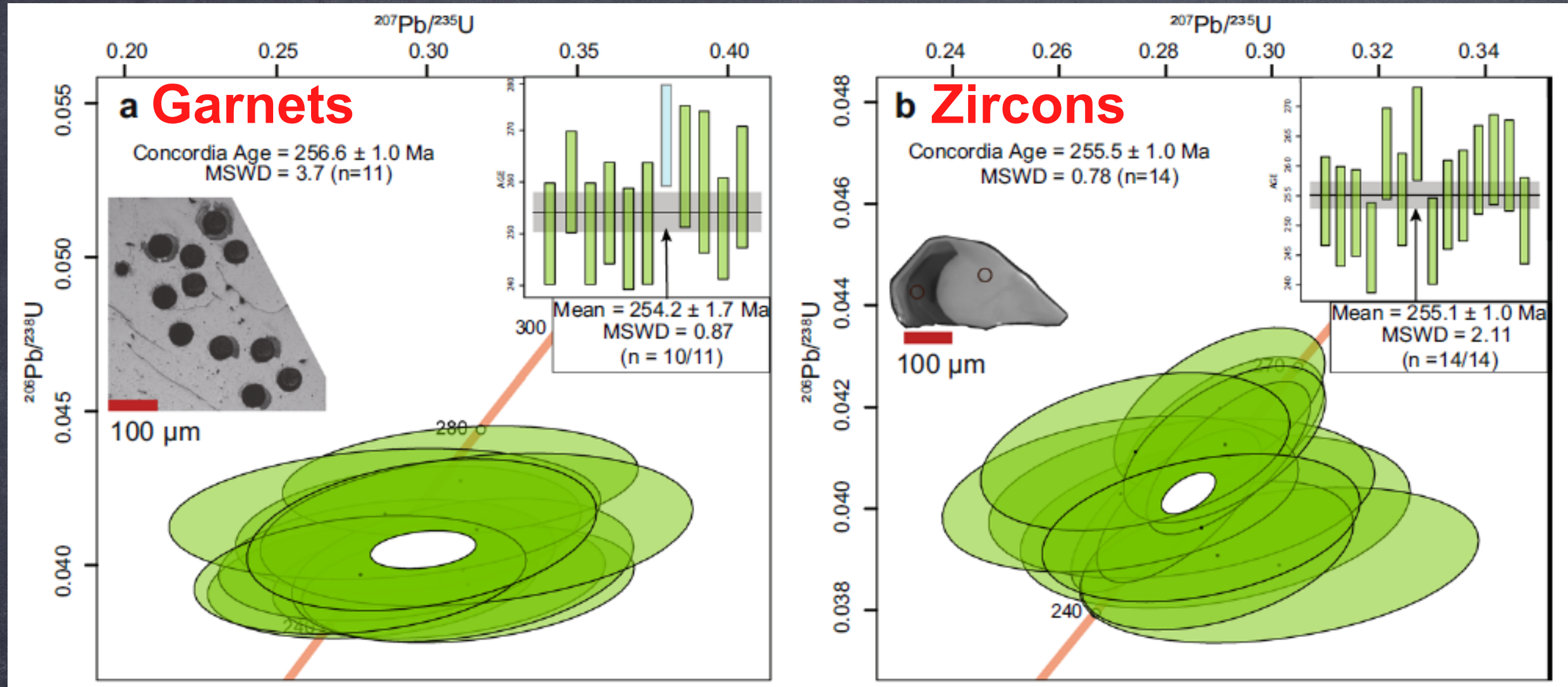


Q. Li <sup>a,\*</sup>, R.R. Parrish <sup>b,c</sup>, M.S.A. Horstwood <sup>c</sup>, J.M. McArthur <sup>a</sup>



**U ranges from 0.47 to 5.3ppm!!!!**

# U-Pb dating: Garnet vs. Zircon



## SCIENTIFIC REPORTS

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Garnet U-Pb and O isotopic determinations reveal a shear-zone induced hydrothermal system

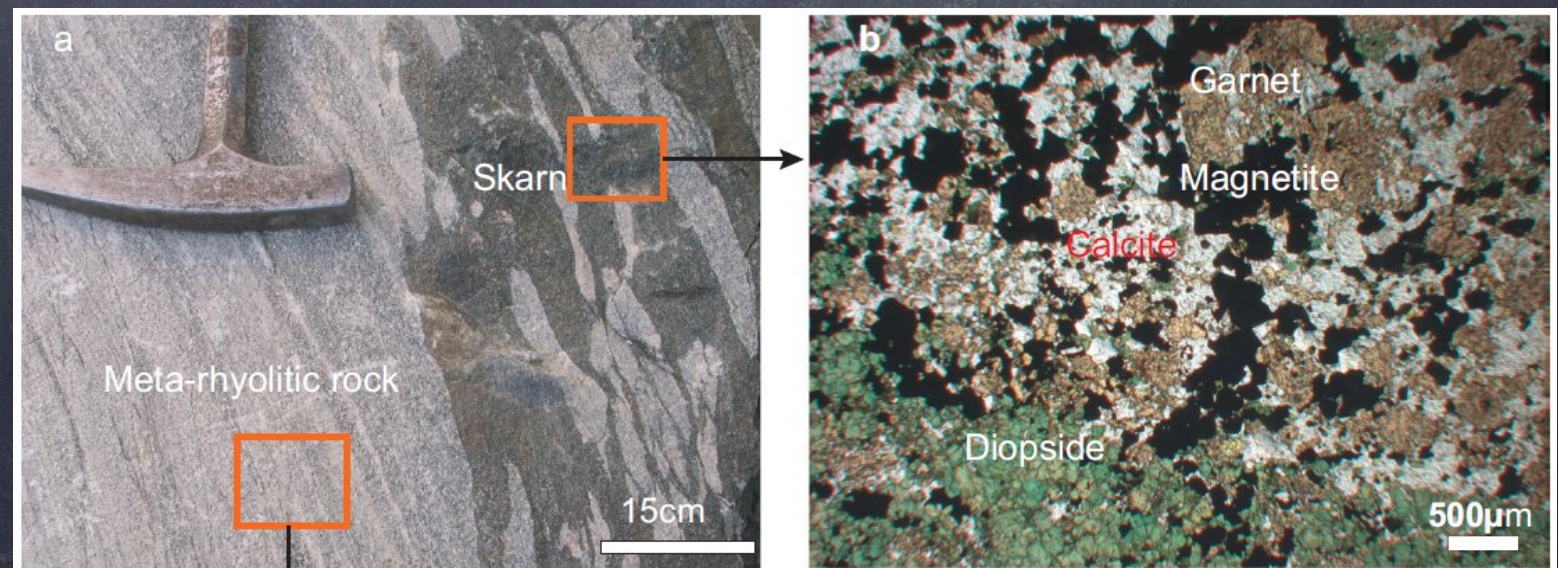
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Andradite:  
 $\text{Ca}_3\text{Fe}^{3+}_3\text{Si}_3\text{O}_{12}$



# U-Pb dating & Traces: Garnet

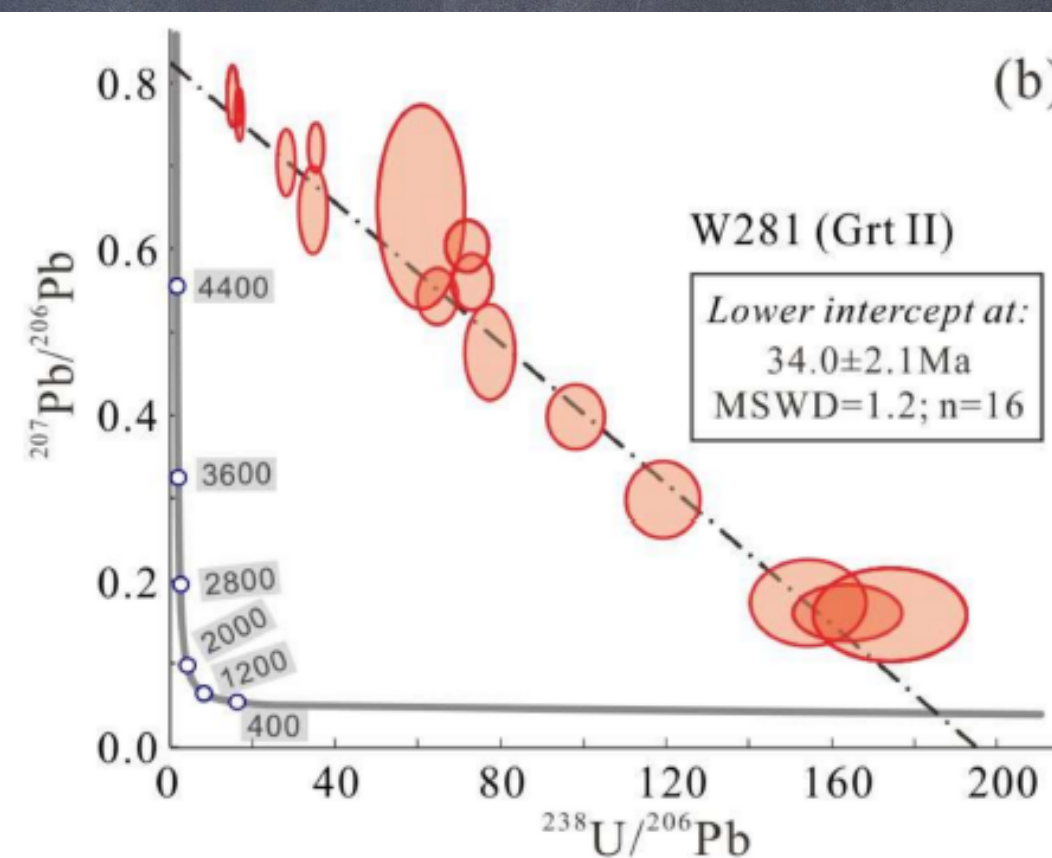
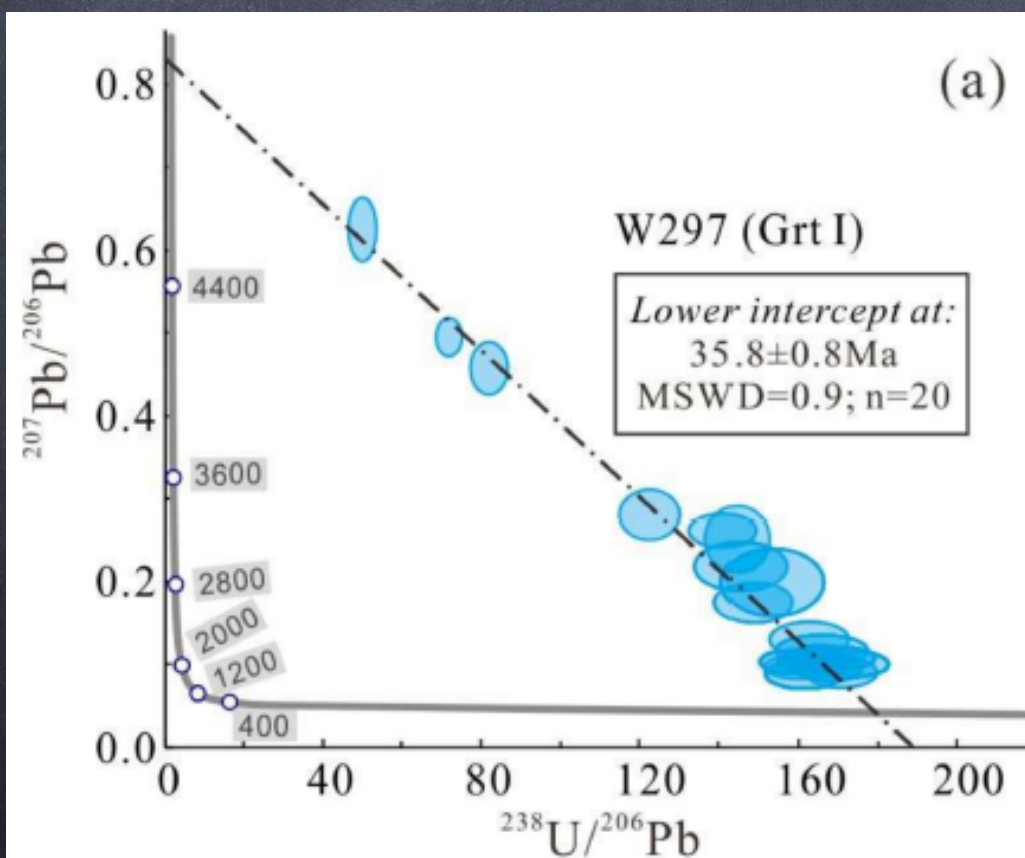
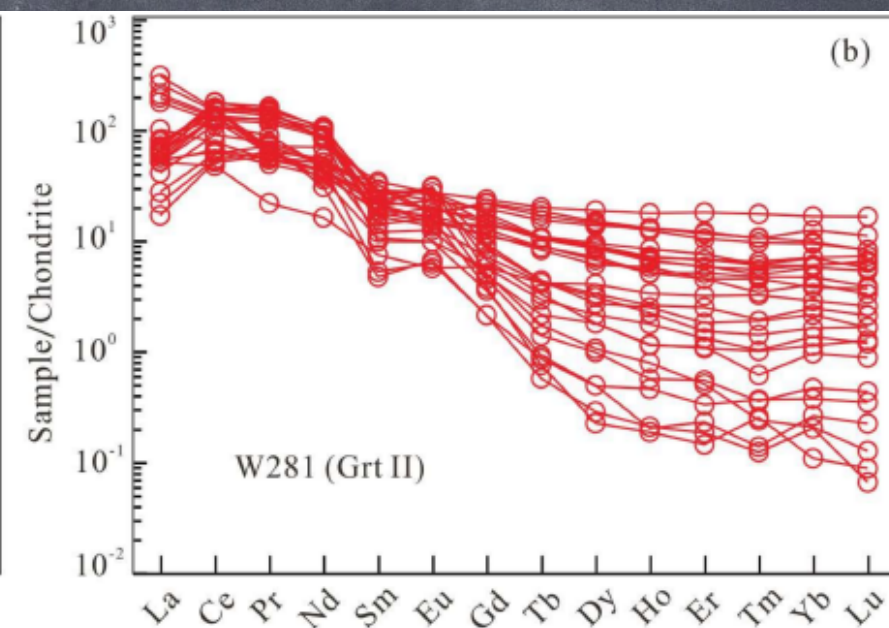
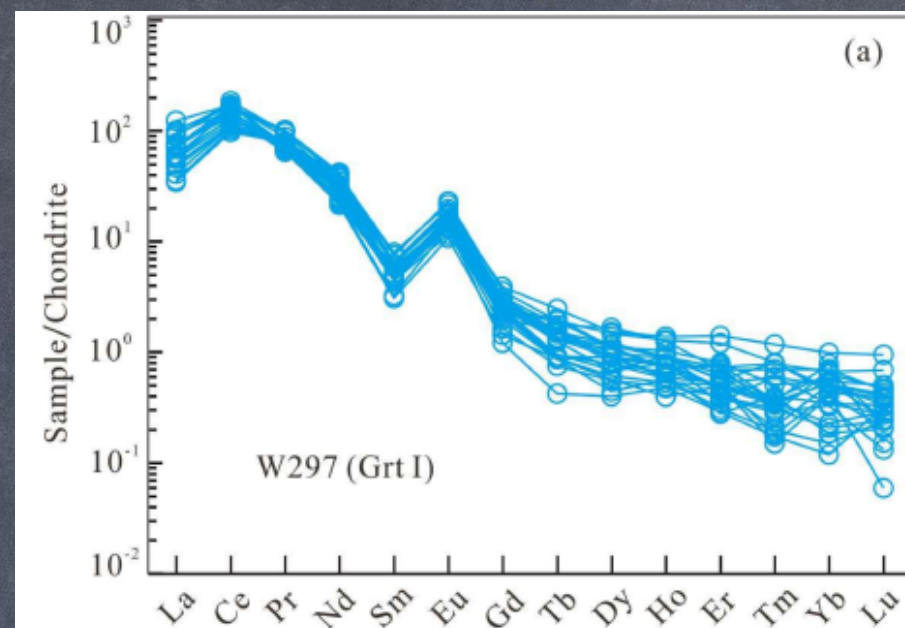
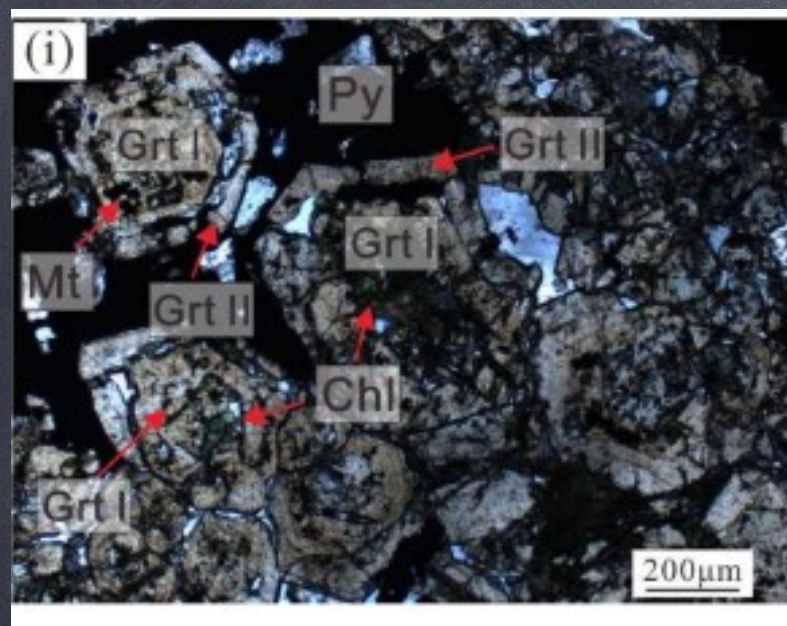


Figure 7. Tera-Wasserburg diagram for the Beiya garnet: (a) W281, (b) W297.

# U-Pb dating: Mantle Garnet

Uranium-lead geochronology applied to pyrope garnet with very low concentrations of uranium

Gary J O'Sullivan<sup>1</sup>, Brendan C Hoare<sup>2</sup>, Chris Mark<sup>1,\*</sup>, Foteini Drakou<sup>3</sup> and Emma L Tomlinson<sup>3</sup>

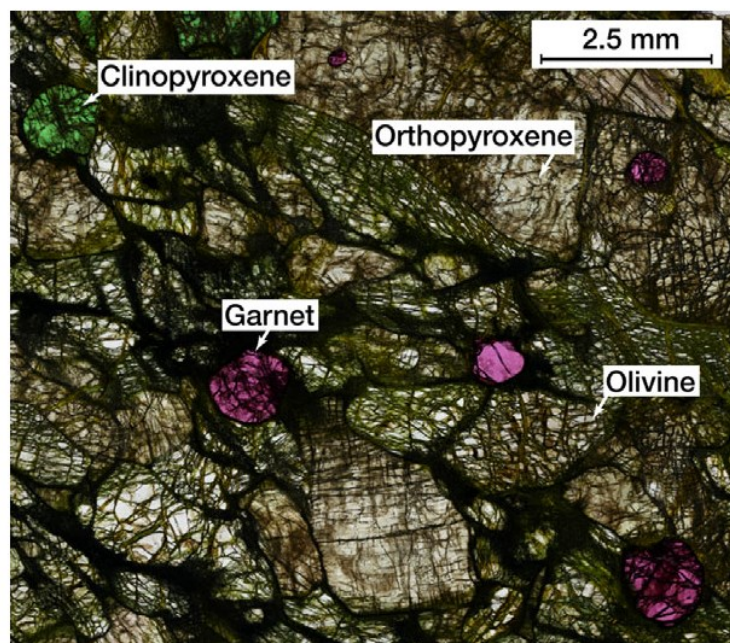


Fig. 2. (Colour online) Transmitted light image of a thick section of coarse-grained lherzolite from Bultfontein (sample CLA-13). Garnet are magenta in colour.

U = 20-85 ng/g  
No Pb comune

ca. 85 Ma

Coincidente con  
età magmatismo

Piropo:  
 $\text{Mg}_3\text{Al}_3\text{Si}_3\text{O}_{12}$

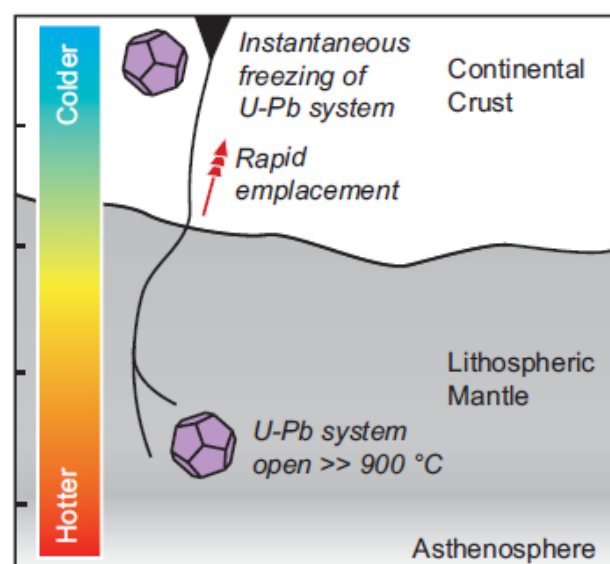


Fig. 5. (Colour online) Conceptual model of the behaviour of the U-Pb system in pyrope-rich garnet from sampled xenoliths. Garnet sampled from very high temperatures, as in this study, will reflect emplacement. Garnet yielding lower equilibration temperatures or encapsulated in diamond (none in this study) might retain ancient information.

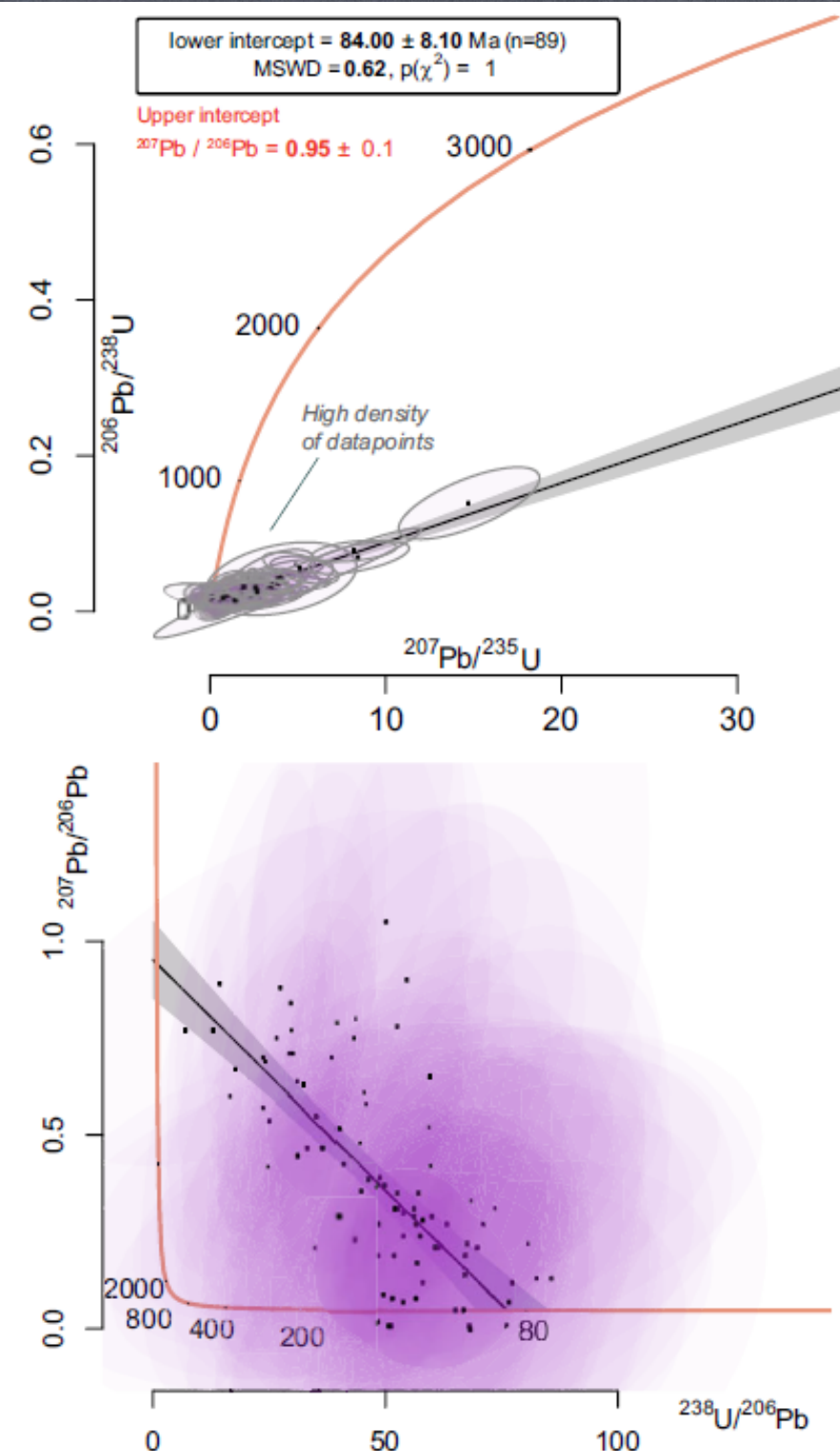


Fig. 4. (Colour online)  $^{207}\text{Pb}/^{206}\text{Pb}$  vs  $^{206}\text{Pb}/^{238}\text{U}$  garnet U-Pb discordia age of Bultfontein xenolith garnet. Age calculations employed the independently measured  $^{207}\text{Pb}/^{206}\text{Pb}$  and  $^{206}\text{Pb}/^{238}\text{U}$  ratios, but are here presented both in a Tera-Wasserburg diagram and transformed for display on Wetherill concordia for comparison. Ellipses are plotted at  $2\sigma$ . In the Tera-Wasserburg diagram, ellipses are plotted with low transparency to show regions of greatest data density.

# Exploring the limits of in-situ U-Pb dating of metamorphic garnet

Aratz Beranoaguirre<sup>1,2</sup>, Leo J. Millonig<sup>2</sup>, Qiao Shu<sup>3</sup>, Richard Albert<sup>2</sup>, Horst R. Marschall<sup>2</sup>, Jose Ignacio Gil Ibarguchi<sup>4</sup>, and Axel Gerdes<sup>2</sup>

Recent advances in analytical techniques and instrumentation allow for the analysis of increasingly smaller sample volumes and lower concentrations. This development significantly expands the possibilities of in-situ geochronology, *e.g.*, LA-MC-ICPMS. Minerals with low U (and Pb) contents such as garnet become the target of in-situ U-Pb geochronology since ages can potentially be obtained from single (sub-)mm-sized garnet grains in thin sections. In this contribution, we explore the current limits of in-situ U-Pb geochronology: what are the minimum concentrations from which an accurate and precise U-Pb age can be obtained?

For that purpose, we have analysed garnets from three different localities that were unsuccessfully analysed in the past using a single-collector sector-field Element XR instrument at FIERCE. These garnets have been re-analysed at FIERCE using a Neptune Plus MC-ICPMS coupled to a RESOLution-LR ArF Excimer laser. The analyses were performed in static mode measuring the masses  $^{206}\text{Pb}$  and  $^{207}\text{Pb}$  with Secondary Electron Multiplier (SEM) and  $^{202}\text{Hg}$ ,  $^{204}\text{Pb}$  and  $^{238}\text{U}$  with the Multiple Ion Counters (MIC). With a spot diameter of 193  $\mu\text{m}$  (round) and a fluence of 2  $\text{J}/\text{cm}^2$  at 15 Hz, ca. 18  $\mu\text{m}$  pit depth was ablated in 18s analysis time, resulting in a total of 2  $\mu\text{g}$  of ablated material. This is more than 2,000 times less material compared to conventional isotope dilution analyses and 3,000 times less U than for a typical LA-ICPMS zircon analysis (20  $\mu\text{m}$  spot). Although the analysed garnets typically have U contents below 10 ng/g, about 15–30 spots are commonly sufficient to define a regression line in the Tera-Wasserburg diagram, yielding a precision of typically <3 % for the lower intercept age. Challenges and details of the method will be discussed using samples of metamorphic garnet from Kaapvaal craton granulites and Eastern and Western Variscan eclogites.

**How to cite:** Beranoaguirre, A., Millonig, L. J., Shu, Q., Albert, R., Marschall, H. R., Gil Ibarguchi, J. I., and Gerdes, A.: Exploring the limits of in-situ U-Pb dating of metamorphic garnet, EGU General Assembly 2022, Vienna, Austria, 23–27 May 2022, EGU22-12133, <https://doi.org/10.5194/egusphere-egu22-12133>, 2022.

- ..with 10 ng/g about 15-30 spots give precision <3%
- This guy is doing U-Pb dating in orthopyroxene...pers. comm

# Laser Ablation Split Stream (LASS) for Zircon Petrochronology

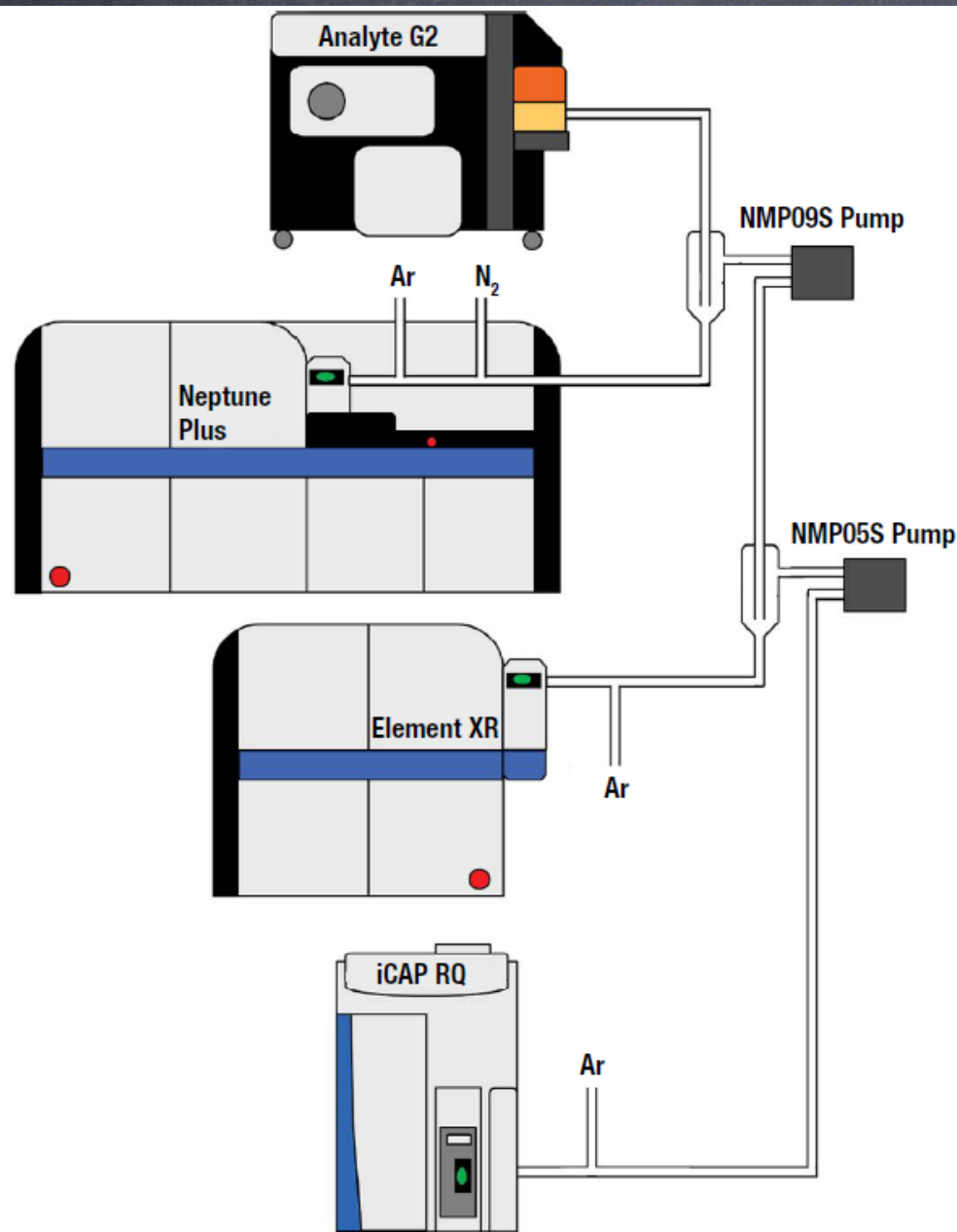


Figure 1. Schematic of LASS to three ICP-MS systems. Approx. 50% of the laser aerosol output was extracted from the primary MC-ICP-MS channel to SF-ICP-MS. Approx. 30% of the SF-ICP-MS channel was extracted to Q-ICP-MS.

GJ-1

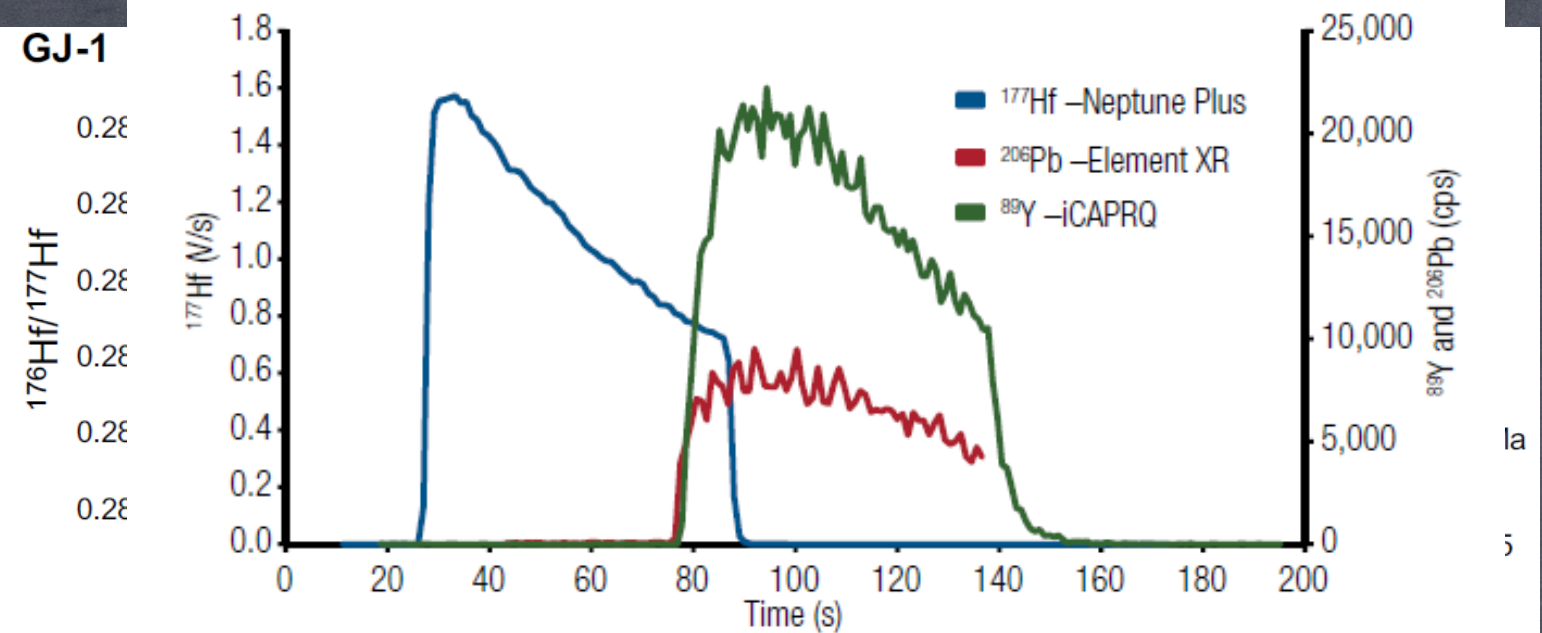


Figure 4.  
Right: U-I

Figure 2. Signal traces for spot ablation of 91500 zircon, LASS to three ICP-MS. Integration times  $\approx 1$  s.  $^{177}\text{Hf}$  (MC-ICP-MS), responded faster than  $^{206}\text{Pb}$  (SF-ICP-MS) and  $^{89}\text{Y}$  (Q-ICP-MS), reflecting the relative distances from the laser ablation cell.

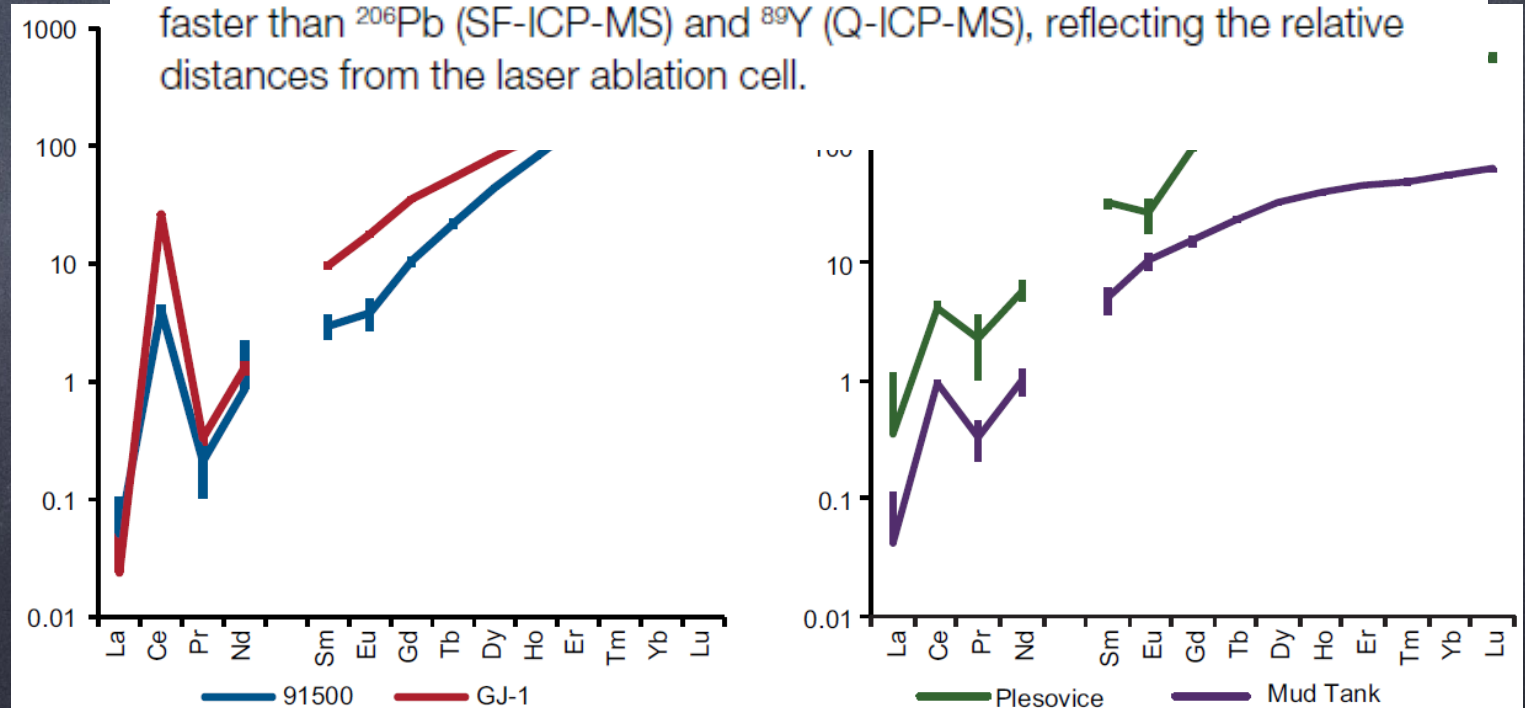


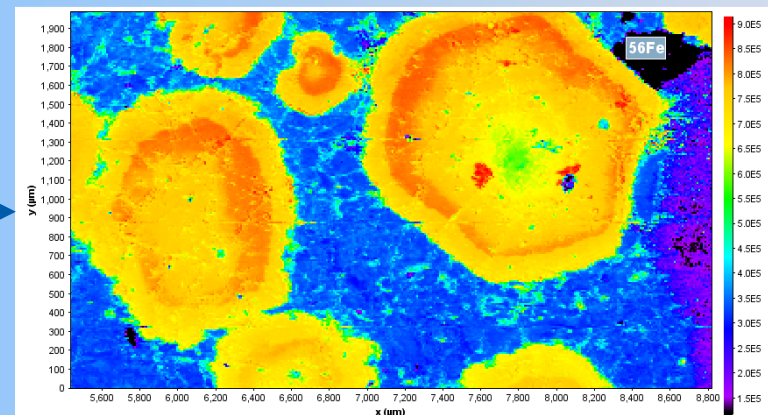
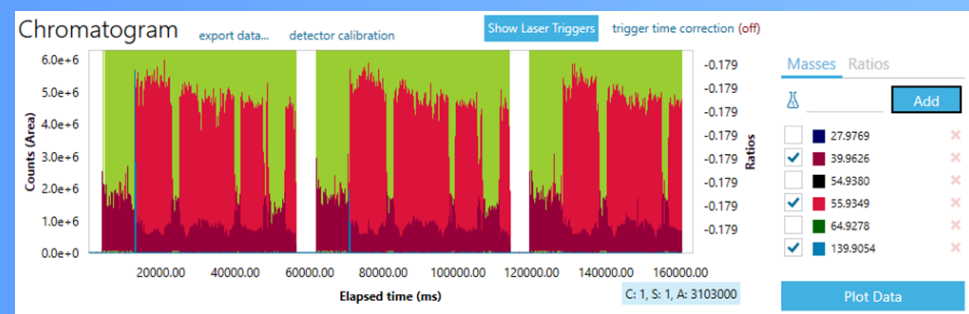
Figure 6. Chondrite-normalized REE concentrations<sup>11</sup> (error bars = SD) determined in the four reference zircons on the iCAP RQ ICP-MS in the LASS configuration.

# ICP-MS-TOF

## ICP-MS Imaging



- 2D elemental map oftentimes used for biological or geological samples
- Instrumental setup: Laser ablation – ICP-MS
  - Sample is shot at with a laser (typically 5-50  $\mu\text{m}$  in size).
  - The generated aerosol is carried to the ICP for analysis
- Modern laser ablation systems can generate separate aerosol clouds every few milliseconds

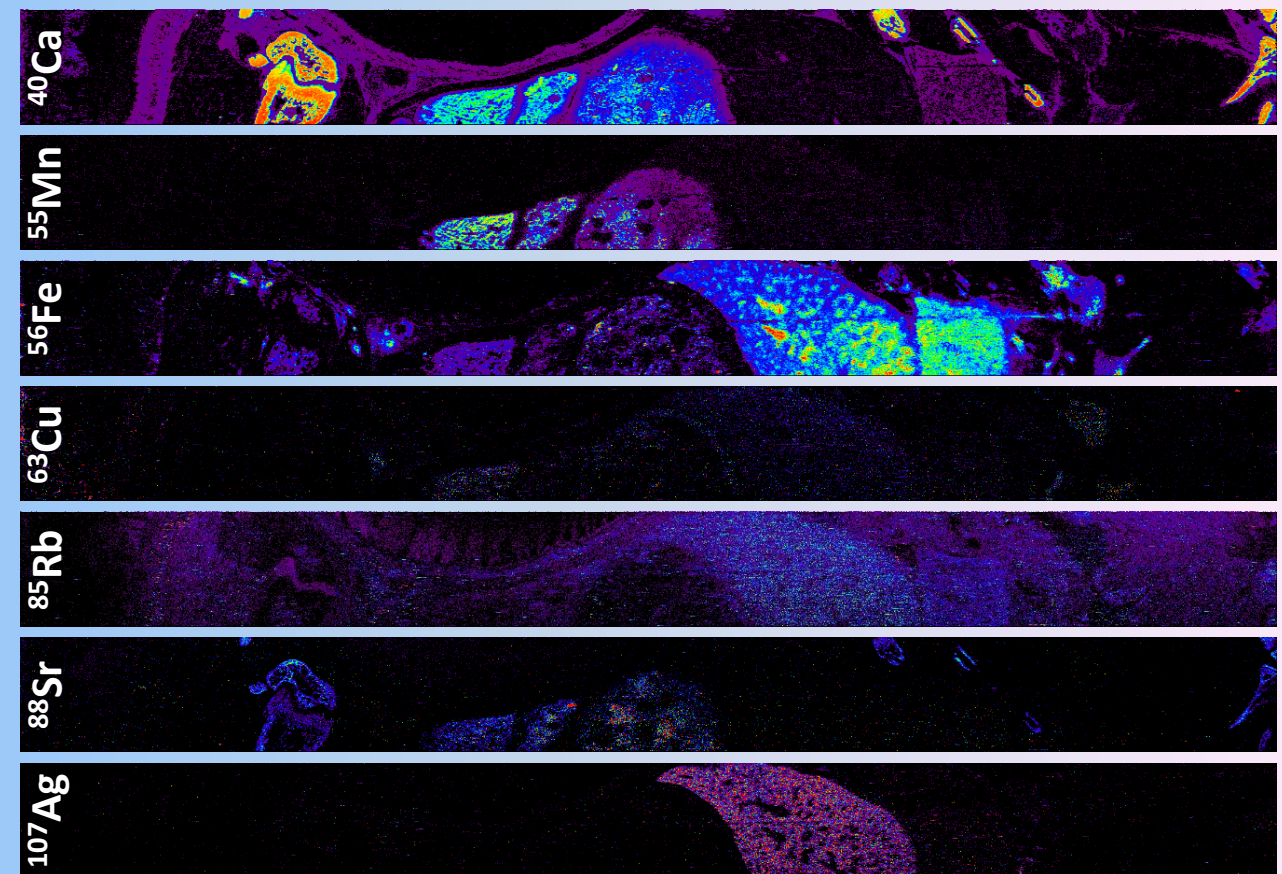
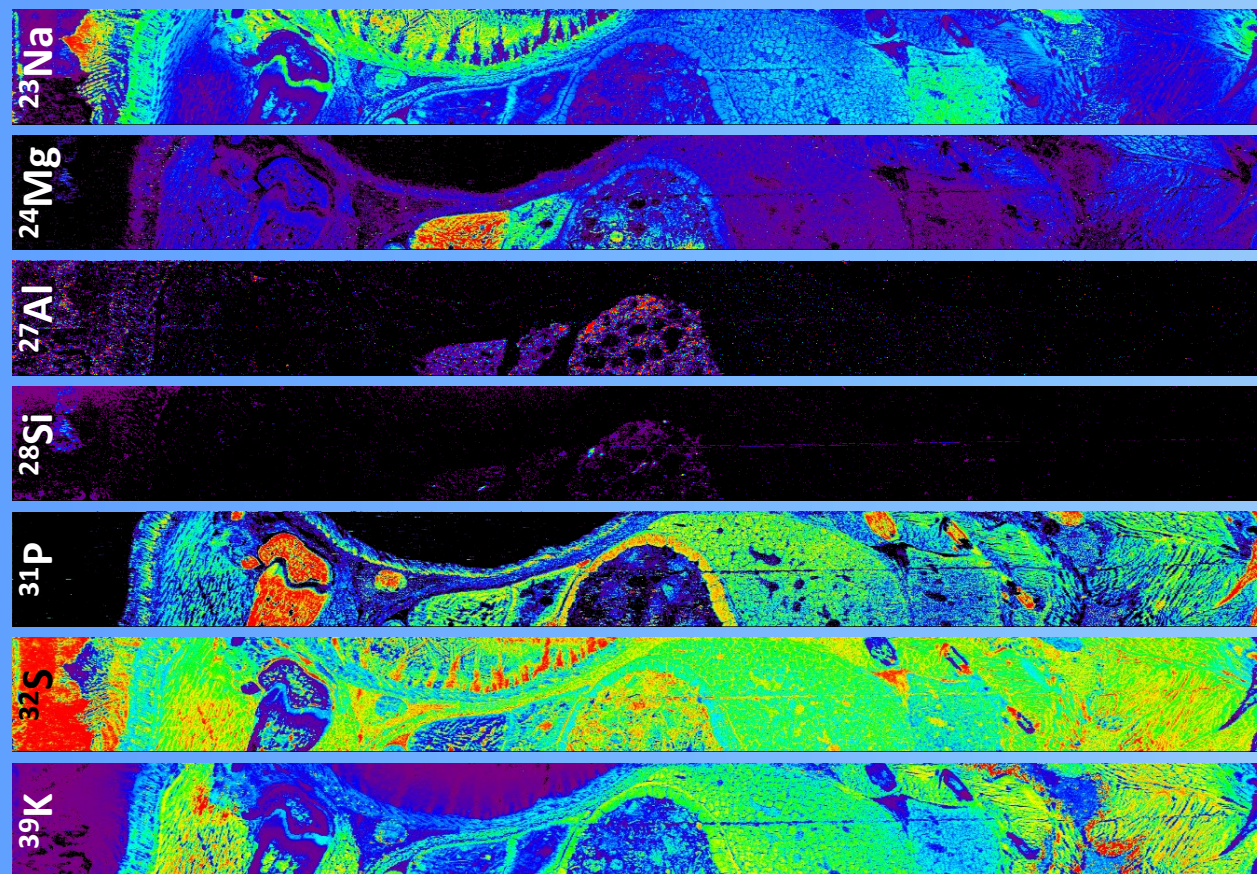


- Combining the coordinates of the laser shot and the intensities results in 2D elemental distribution map
- Pixels can only be resolved if every laser shot is discriminated

# ICP-MS-TOF

## Multi-element imaging

- Elemental maps of the entire periodic table
  - Thin section of entire mouse*

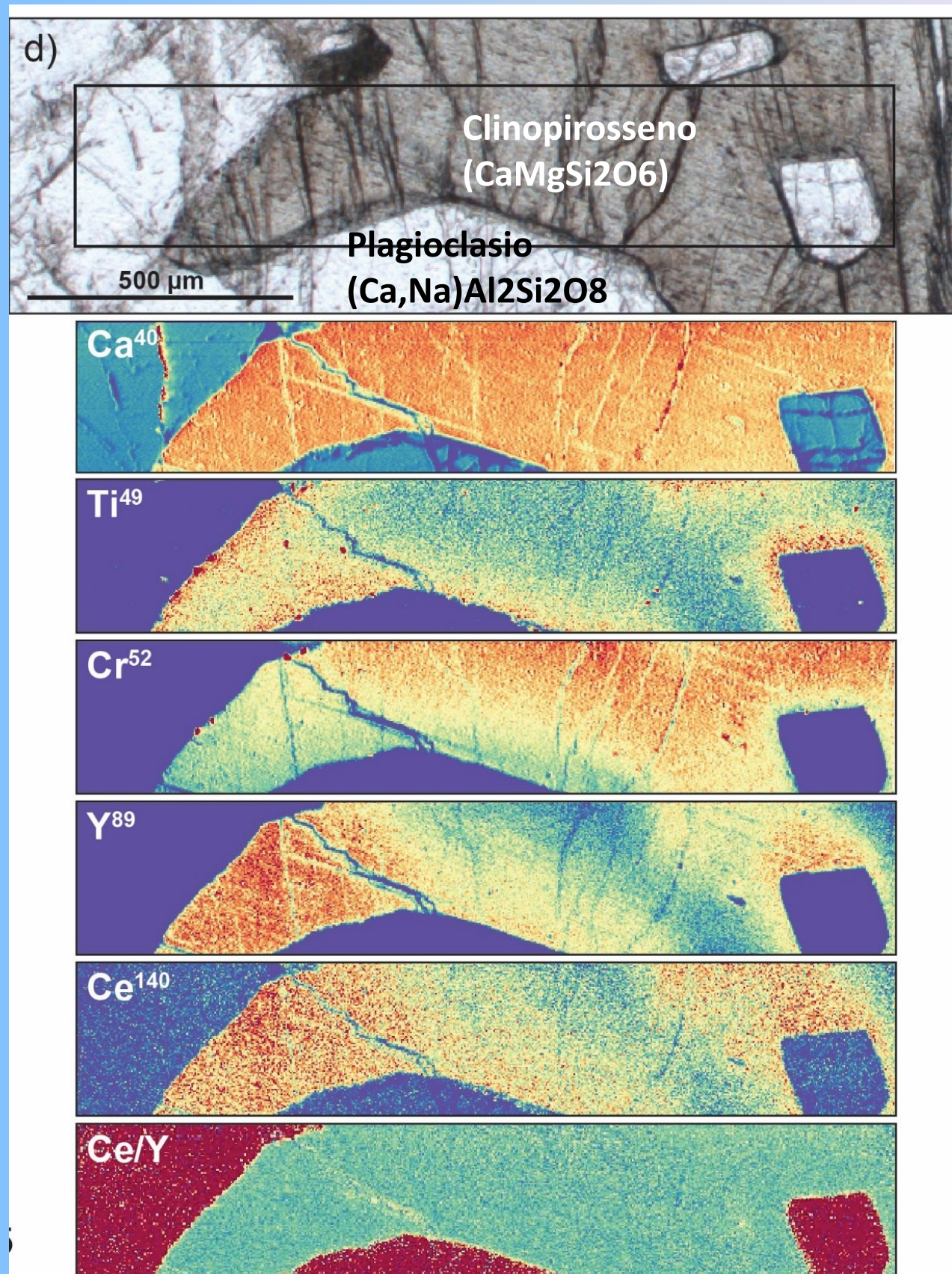


# ICP-MS-TOF

Gabbro oceanico da  
Hess Deep, che  
rappresentano i  
cumulati più impoveriti  
in Terre Rare di  
ambiente oceanico

Basch et al. (2024)  
Journal of Petrology

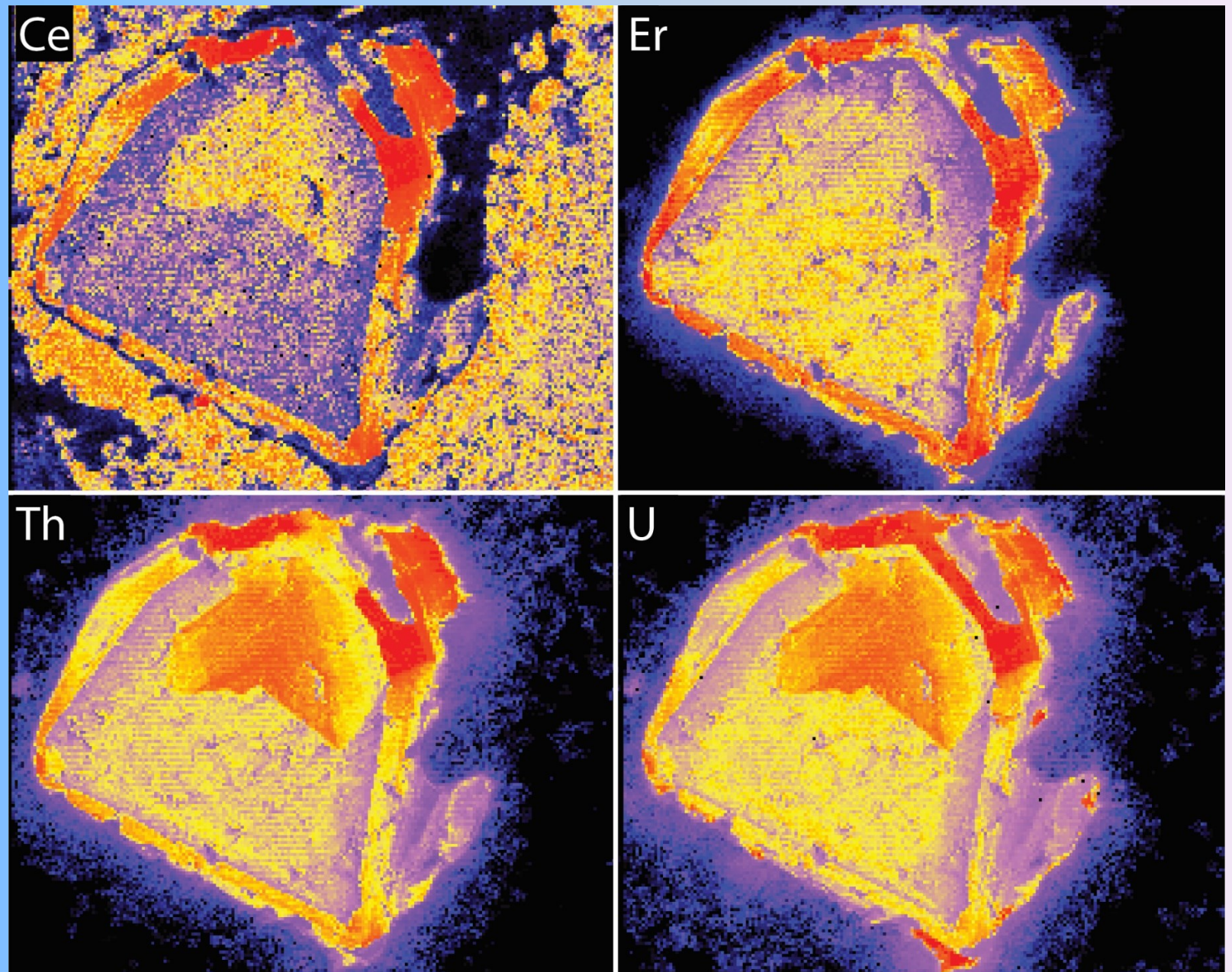
ArF Excimer ESI + Vitesse (Nu)



# ICP-MS-TOF

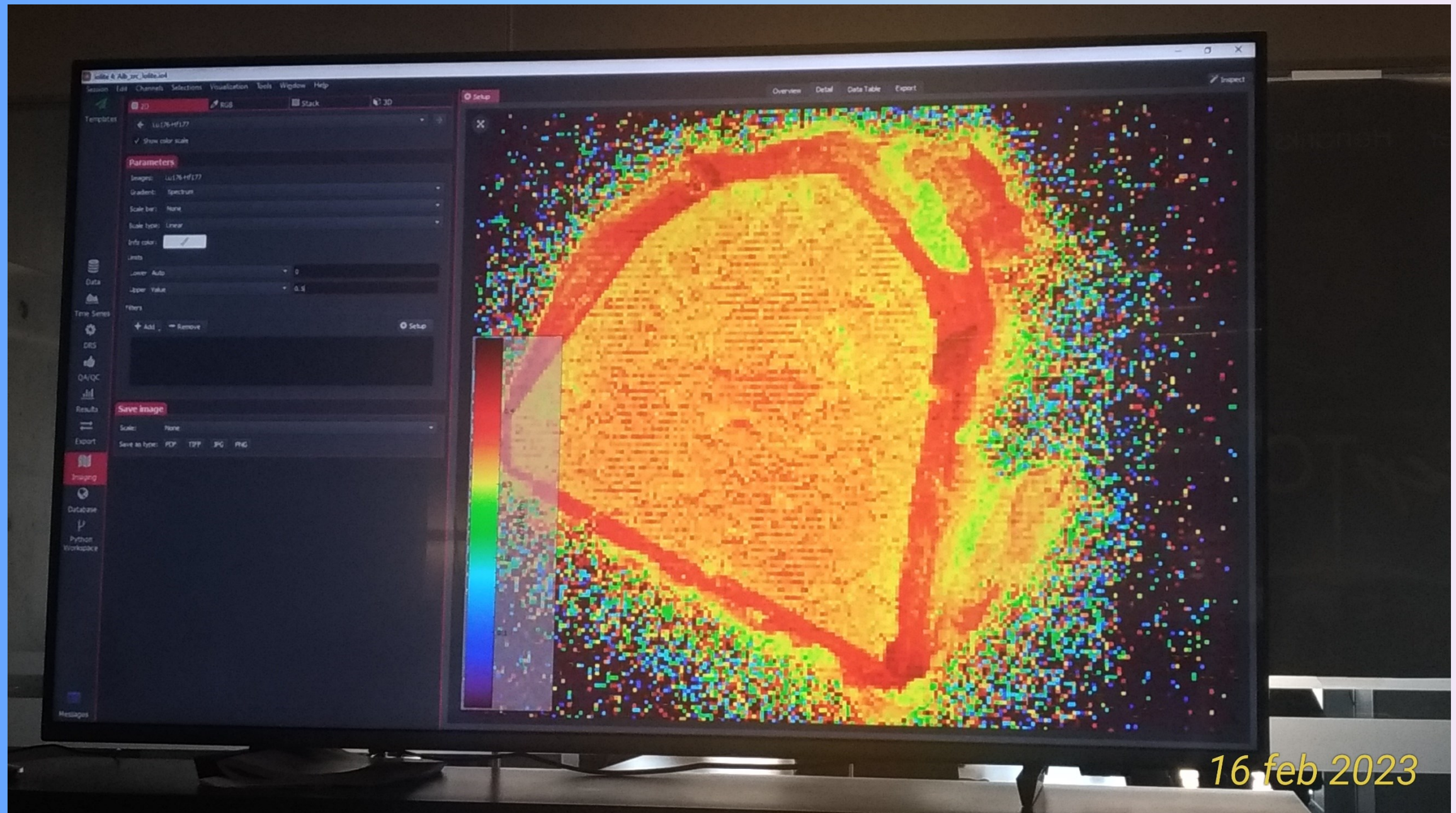
Elementi in tracce in  
Zircone Magmatico

ArF Excimer ESI +  
TofWerk



1 mm

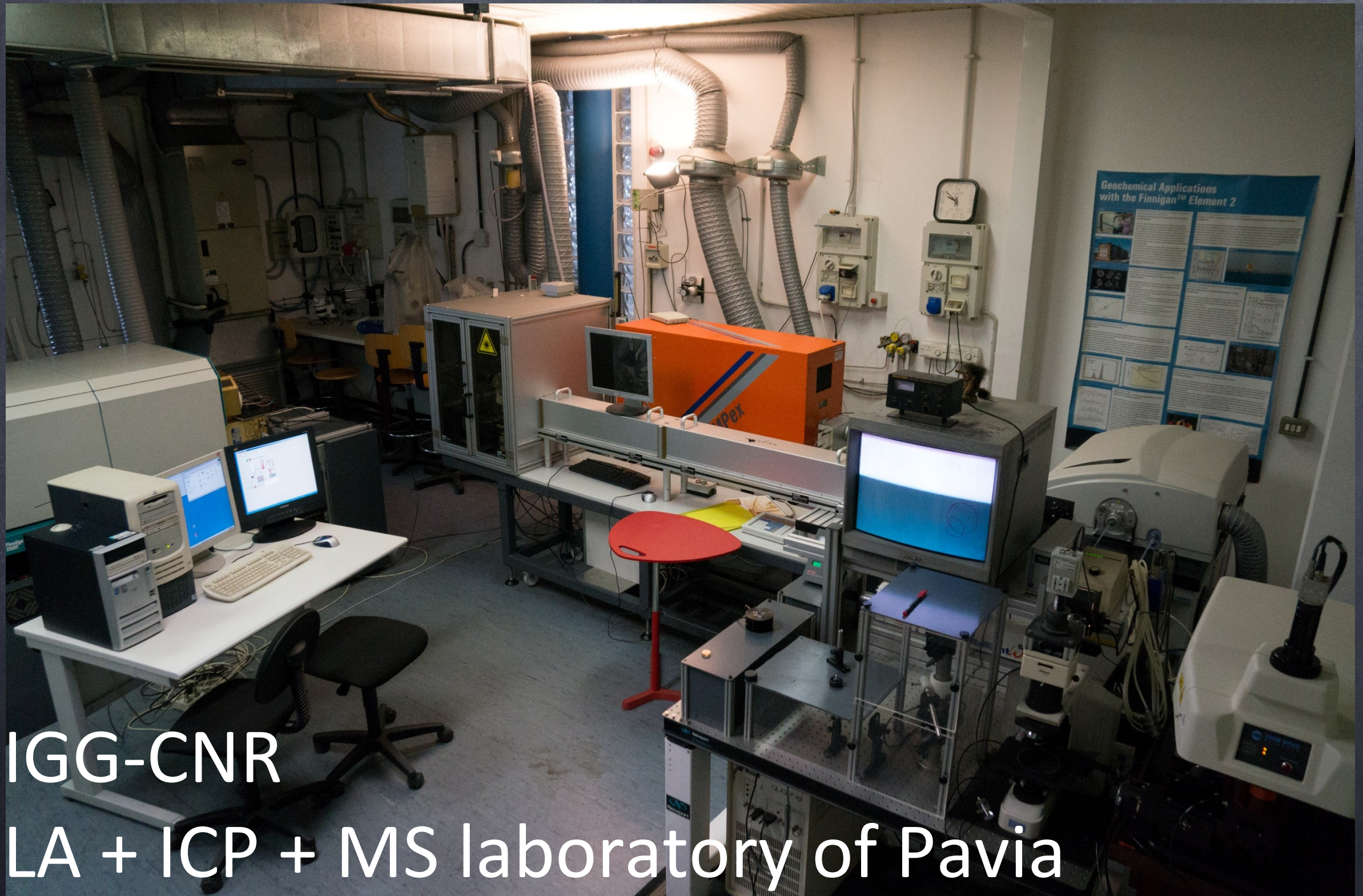
# ICP-MS-TOF



1 mm

Rapporto isotopico  $\text{Lu}^{176}/\text{Hf}^{177}$  in Zircone magmatico

# Grazie



IGG-CNR  
LA + ICP + MS laboratory of Pavia

# ISTITUTO DI GEOSCIENZE E GEORISORSE (IGG)

NEPTUNE-NEOMA LABORATORY

LAB HEAD: DR SAMUELE AGOSTINI



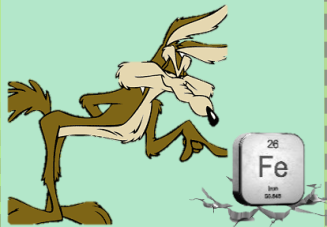
AT PRESENT

NEW  
SYSTEMATIC

IN PROGRESS

IN PROGRAM

| Rb-Sr                               | Sm-Nd                                                      | Pb-U-Th                             | B                                   | Fe                       | Li                        | U-Th DATING OF SPELEOTHEMS              |
|-------------------------------------|------------------------------------------------------------|-------------------------------------|-------------------------------------|--------------------------|---------------------------|-----------------------------------------|
| TRACING MIXING OF GLOBAL RESERVOIRS | ISOTOPE MARINE SIGNATURE AND PALAEOCEANOGRAPHIC VARIATIONS | EVOLUTION OF SOLAR SYSTEM           | MIXING AND TRACING PROCESSES        | METAL SOURCE             |                           |                                         |
| HIGH-T PROCESSES                    |                                                            | FORMATION OF EARTH CRUST            | CONTINENTAL WEATHERING              | HYDROTHERMAL CIRCULATION | LI-RICH GEOTHERMAL FLUIDS | IN SITU B ISOTOPES ON CARBONATE SAMPLES |
| WATER-ROCKS WEATHERING              | MANTLE AND CRUST-DERIVED IGNEOUS CRUST DISCRIMINATION      | ORE-FORMING MATERIAL SOURCES        | ANCIENT MARINE PH RECONSTRUCTION    | ORE-FORMING PROCESSES    | HIGH-T PROCESSES          | IN SITU Rb-Sr DATING                    |
| DATING CARBONATE SUCCESSIONS        |                                                            | GEOCHRONOMETER                      | MAGMATIC ACTIVITY                   | CRITICAL ZONE STUDIES    | WATER-ROCK INTERACTIONS   | Mg SYSTEMATICS                          |
| OCEANIC WATER MIXING                |                                                            | GEOGENIC OR ANTHROPOGENIC POLLUTION | WATER-ROCKS WEATHERING              | HIGH-T PROCESSES         |                           | Zn-Cu SYSTEMATICS                       |
|                                     |                                                            |                                     | GEOGENIC OR ANTHROPOGENIC POLLUTION | HUMAN HEALTH             |                           | Ca SYSTEMATICS                          |
|                                     |                                                            |                                     | ORE-FORMING PROCESSES               |                          |                           |                                         |





## **Cella porta campioni fast-washout del sistema LASER**

**Da ablazione a analisi  
pochi ms**

**Analisi di qualunque  
matrice solida, geo e bio**



# LA-ICP-MS Trace element

## No gas MS/MS mode

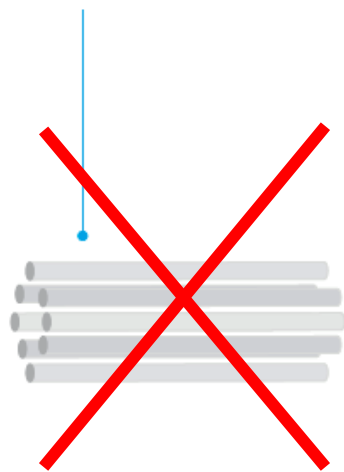
### ICP-MS a triplo quadrupolo Agilent con MS/MS

**Filtro di massa quadrupolare (Q1)**  
Scarta tutti gli ioni fuori massa prima che possano entrare nella cella



Solo l'analita e le interferenze sulla massa raggiungono la cella

**Cella di reazione**  
Usa la chimica di reazione per risolvere l'analita dalle interferenze sulla massa

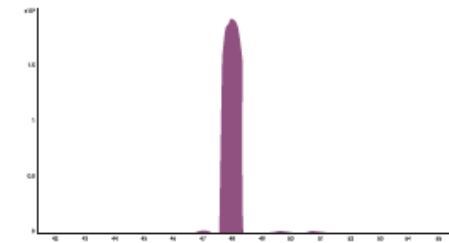


Solo gli ioni dell'analita (e gli ioni fuori massa/ioni prodotto) escono dalla cella

**Filtro di massa quadrupolare (Q2)**  
Scarta gli ioni fuori massa e lascia passare al rivelatore l'analita senza interferenze



Solo gli ioni dell'analita target o gli ioni prodotto contribuiscono al segnale misurato, cosicché i risultati sono accurati e coerenti



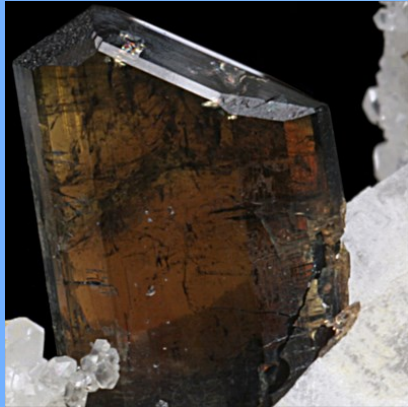
MS/MS improves the accuracy and confidence, reduce the oxides

.... to correct isobaric interference

Titanite



Allanite

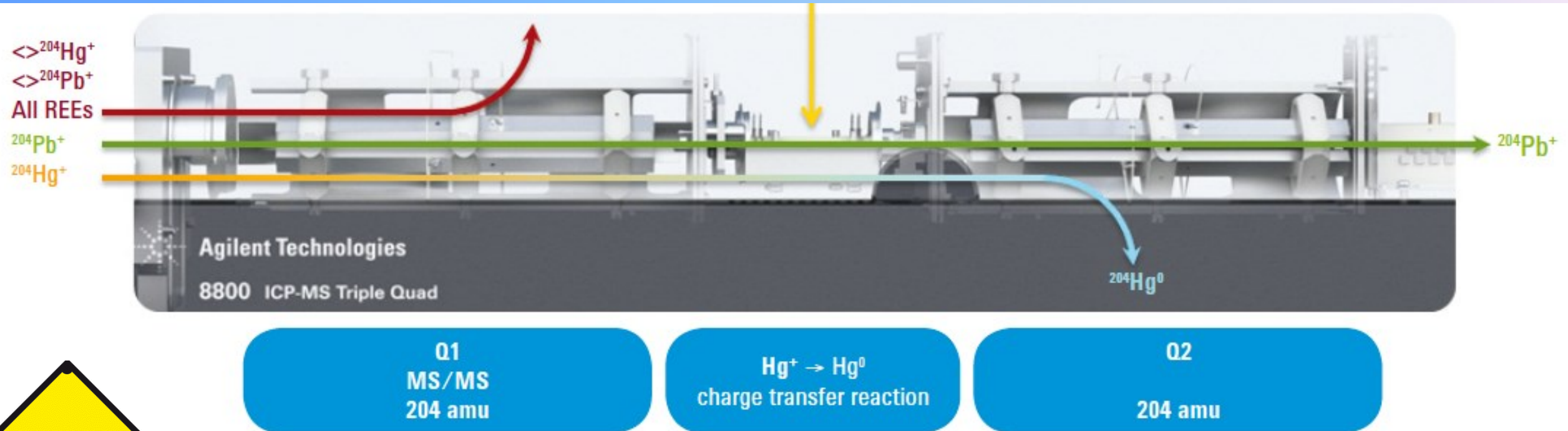


contain Hg =>

$^{204}\text{Hg}$ - $^{204}\text{Pb}$  isobaric interference

**Lead isotope analysis: Removal of  $^{204}\text{Hg}$  isobaric interference from  $^{204}\text{Pb}$  using ICP-QQQ in MS/MS mode**

**$\text{NH}_3$  reaction gas**



Schematic representation of MS/MS reaction mode for Hg interference removal. Q1 is set to  $m/z$  204 rejecting all ions at masses other than 204 amu

# QQQ: what we would like to do

## Rb-Sr dating

### (NO<sub>2</sub> in reaction cell)



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Procedia Earth and Planetary Science 17 (2017) 464 – 467

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Earth and Planetary Science

15th Water-Rock Interaction International Symposium, WRI-15

In situ Rb-Sr dating of fine-grained vein mineralizations using LA-  
ICP-MS

Mikael Tillberg<sup>a,1</sup>, Henrik Drake<sup>a</sup>, Thomas Zack<sup>b</sup>, Johan Hogmalm<sup>b</sup>, Mats Åström<sup>a</sup>

